

# Advancement in the sulfone-based dendrimers: From synthesis to application

Shaziya Khanam, Sunil K. Rai, Ashish K. Tewari\*

Department of Chemistry (Center of Advanced Study), Institute of Science, Banaras Hindu University, Varanasi-221005, Uttar Pradesh, India

\*Corresponding author: E-mail: ashishtewarichem@gmail.com

Received: 08 November 2016, Revised: 29 November 2016 and Accepted: 13 December 2016

DOI: 10.5185/amlett.2017.1609

www.vbripress.com/aml

## Abstract

The development of sulfone-based macromolecules (polymer to dendrimers) has been inspired and prompted by much interesting importance of sulfone chemistry. Literature reveals that dendrimers are more demanding than the polymer in many aspects, because of its unique structural properties. In spite of increasing importance and applications of dendrimers, it has a lot of hindrance regarding the tedious way of synthesis, which requires large excess of reagent and several problems in isolation. Therefore, time to time several tactics of synthesis has been developed. It was also well known that sulfonylation reaction is not only fast and easy to handle but also aryl sulfonylchloride derivatives are inexpensively available. Nowadays, sulfone-based dendrimers are exponentially expanding its application in disciplines of science which lie in organic synthesis, material sciences, combinatorial chemistry, medicinal chemistry and chemical biology. In this review, we have compiled the types of dendrimer and chiefly focused on the synthesis of sulfone-based dendrimers and their applications, in short. Further, future prospect of sulfone-based dendrimers in various branches of science has been outlined. Copyright © 2017 VBRI Press.

**Keywords:** Dendrimer, generation, sulfone, sulfonimide, sulphonamide, suzuki cross coupling.

## Introduction

Advancement of organic chemistry has given a lot of method of synthesis (solvent free to metal catalyzed, low temperature to high temperature and photochemical to microwave reaction conditions). These methods have been extensively utilized in a broad range of research areas including natural product synthesis [1,2], material science [3], medicinal [4], biological [5] supramolecular chemistry [6], catalysis [7], coordination chemistry [8] and macromolecule synthesis [9-10]. Moreover, some of the reactions (Suzuki-Miyaura [11], Heck [12], Staudinger [13], Ullman [14,15], Sonogashira [16,17], and click reaction (cycloaddition reaction) [18-20]) are very powerful tools for the creation of smart macromolecules i.e. branched dendritic molecules [21], which are now become an exciting area of research. Branched dendritic family includes dendrimers, dendrons, dendrigraft and hyperbranched or HY polymers (Fig. 1) [22,23].

Among branched macromolecules, dendrimers represent a key component because of its unique properties like monodispersity, globularity and multiple end-groups at the periphery of the dendritic framework, which does not show by other branched macromolecules and small molecules. Thus dendrimers are foreseen to deliver extraordinary features, for applications in areas such as cancer therapy, biosensors for diagnostics and light harvesting scaffolds, compared to small and

branched macromolecules of the same molecular weight [24]. Inspired by these novel features of dendrimers, it attracted a lot to the polymer scientists to focusing on generating new chemistries that can deliver monodispersed polymers with structural perfections and enhanced functions.

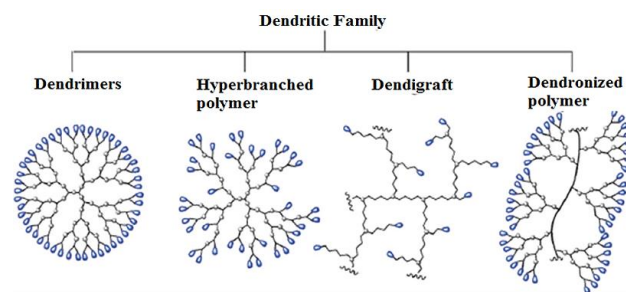


Fig.1: Structural representation of the dendritic family.

Flory's, 1941 opened the gate for the synthesis of branched three-dimensional structures [25, 26]. The synthetic work of such extraordinary molecules was first pioneered by Vögtle in 1978 and termed it "cascade molecules" [27]. A few years later, in the early 1980s, Denkwalter et al. patented the synthesis of L-lysine-based dendrimers upto high generations, but they had not given a complete characterization of the dendrimers except for size exclusion chromatography (SEC) data [28]. The properties of dendrimers were first examined by

Tomalia's PAMAM dendrimers and Newkome's arborols [29-31] (unimolecular micelles), then with Meijer's poly(propylene imine) (PPI dendrimers) [32]. Now PAMAM and PPI become most studied and commercialized dendrimers. Nowadays, in the field of synthetic organic chemistry dendrimers become one of the most exciting structures to synthesize and access in comparison to their linear analogs. The field of dendrimers has been continuously reviewed [33-43]. Since an excessive number of dendrimers have continuously been reported. Therefore, we shall shortly describe the dendrimer by different type.

### Types of dendrimers

Dendrimer can be divided on the basis of:

1. Structural design: This category of dendrimer have been constructed by their shape, i.e., spherical [44], dumb-bell [45,46], tadpole shape [47,48], snowflake shape [49], multilingual [50] and Janus [51] as shown in Fig. 2.
2. A way of synthesis: This category of dendrimers are classified by their synthetic strategies. The synthetic strategy was started with convergent [33] and divergent [52] method. Now, these come under the conventional growth approach [34]. Further, several strategies had been developed, which came to the accelerated growth approach [53]. The accelerated growth strategies have been divided into two parts as demonstrated in Fig. 5. We briefly explain these strategies.

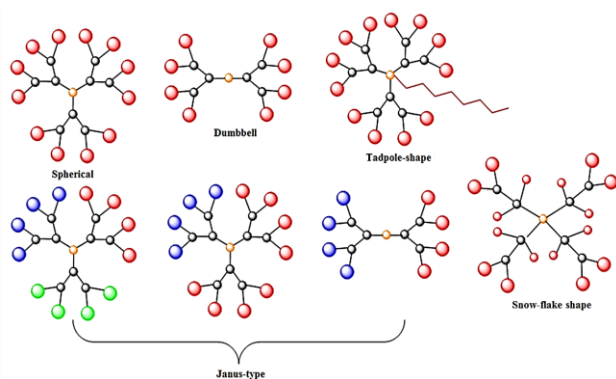


Fig. 2. Different shape of dendrimers.

### Conventional synthetic strategy

(i) Divergent method: In this approach synthesis starts at the core proceeded outwardly to the periphery of the dendrimer. The basic strategy has been demonstrated by cartoon diagram as shown in Fig. 3. The process starts with the coupling between site C and focal point F leading to the intermediate with protected periphery indicated by the letter P. Activation of P at the periphery gave another intermediate with activated periphery indicated by the letter C. Further, coupling with focal point F of the another branched monomer forming upto second generation dendrimer possessing protected functional group P, which becomes a convenient place for a new

active site to react with new branched units to increase generation in dendritic framework.

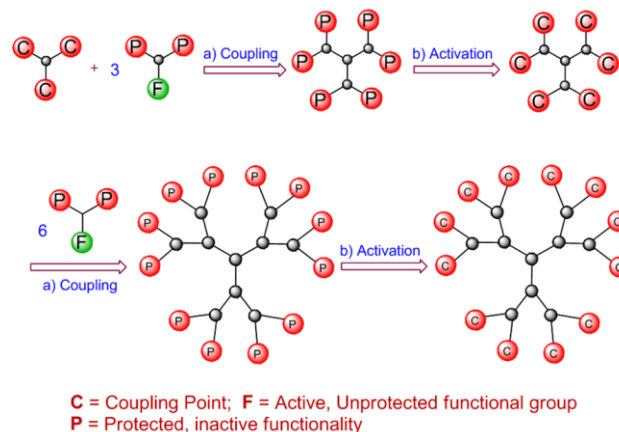


Fig. 3. Cartoon diagram is representing divergent synthesis.

(ii) Convergent method: In this approach synthesis started at the periphery and proceeded inwardly to the core. It operates inwardly by increasing generation in branched monomer followed by activation and deactivation steps and attached to a core as shown in Fig. 4. Through this method, obtained dendrimers are highly pure and monodispersed.

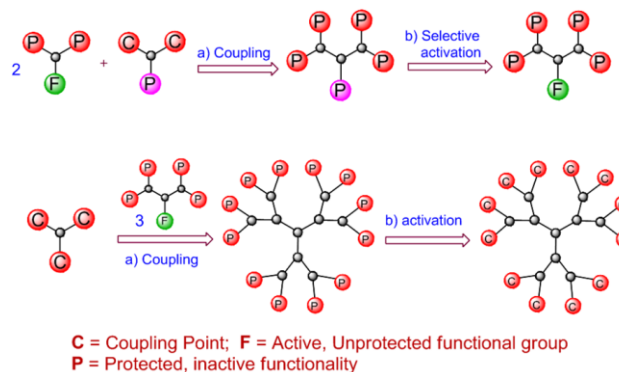


Fig. 4. Cartoon diagram is representing convergent synthesis.

Above two methods have advantage and disadvantage too. Therefore, several other methods have also been developed which are either mixture of the convergent and divergent or some modification in them. Now, these modified methods are called revised traditional strategy. These approaches increase the number of branching in the dendritic framework by decreasing number of synthetic steps but could not eliminate activation and deactivation steps. Several dendrimers have been synthesized using conventional or revised traditional strategies [54-57].

These are as follows:

### Revised traditional strategies

(i) Double-stage convergent growth approach: It combines both convergent and divergent approach and provides new alternative pathway of synthesis, in which both hypercore and dendrons were synthesized parallel through activation and deactivation steps. Further, focal points of dendrons are coupled in a divergent manner to

the periphery of a hypercore, prepared either by convergent or divergent growth. The synthetic strategy has been shown in Fig. 6. The first successful attempt regarding this approach was made by Wooley [58] et al.

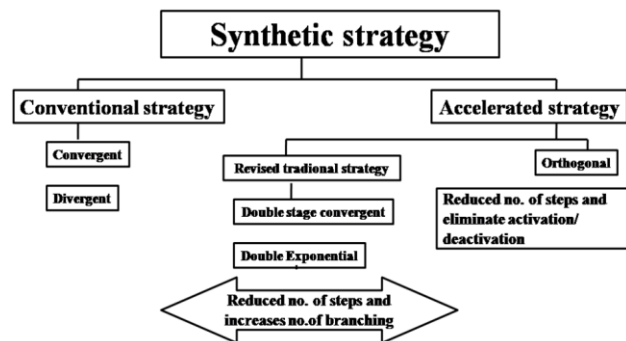


Fig. 5. The flow diagram is showing the synthetic strategies for dendrimers.

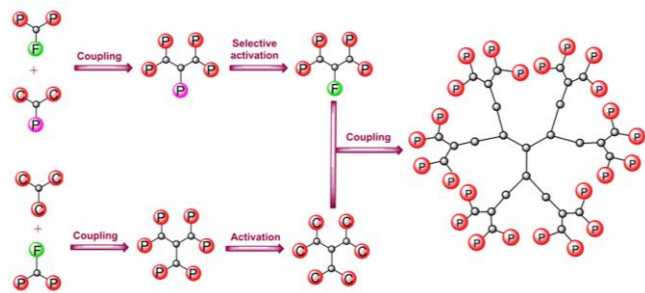


Fig. 6. Cartoon diagram is representing double-stage convergent synthesis.

(ii) Double-exponential dendrimer growth: This approach has been initiated by Moore et al. and developed by Hult *et al.* [59, 60]. The strategy has been demonstrated in Fig. 7. Two kinds of small branched dendron (activated/deactivated) were utilized. The coupling of these small dendrons was carried out by making activation in them. One dendron is activated selectively, at their focal point while other at the periphery and then coupled together to obtain a higher generation fully protected dendron. These steps were repeated to increase dendrons generation and, in the final step, the focal points of the dendrons are activated and coupled to either tri-core or multi-functional core. Thus, The high molecular weight dendrimers were constructed in a lesser number of steps.

So far, scientists have developed newest and simpler routes to construct dendrimers without activation/deactivation steps. Now, these strategy comes under the orthogonal strategy which is accelerated strategy. Further, orthogonal growth concept will be discuss.

#### Orthogonal coupling strategy

Recently, Zeng and Zimmerman developed an alternative method, that is short period synthesis of the dendrimer, to overcome the difficulties of revised traditional strategy (double stage convergent and double exponential growth approaches) [60, 61]. In the orthogonal methodology, no

activation step is required. Two different kinds of dendrons having different functionality ( $AB_2$  and  $CD_2$ ) were utilized for increasing branching units.

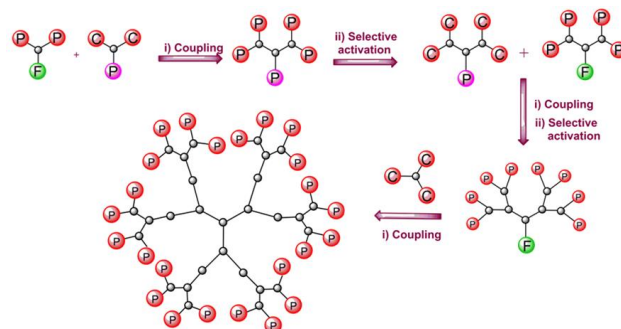


Fig. 6. Cartoon diagram is representing Double-exponential convergent synthesis.

The functionalities at periphery and focal points should be arranged in such a manner that  $AB_2$  dendron couple to the site C, and surface functionalities B should be neutral toward C. Similarly, when  $CD_2$  dendron couples to the site B, surface functionalities D should be neutral toward B. Alternatively,  $AB_2$  and  $CD_2$  should be coupled to the desired core (Fig. 7). Thus activation and deactivation step are eliminated.

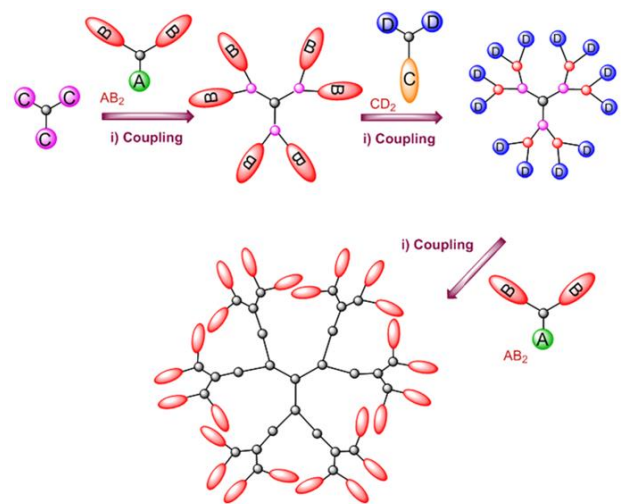


Fig. 7. Cartoon diagram is representing the orthogonal synthesis.

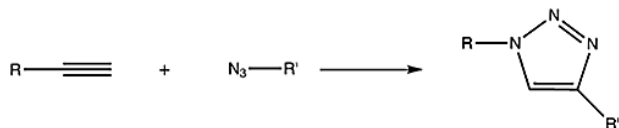
i. Chemoselective strategy: By orthogonal concept, Barany and Merrifield (in 1977) introduced the concept of chemoselectivity. This strategy has been utilized for peptide synthesis. The strategy was utilized for the construction of dendritic framework, which was first initiated by Spindler and Fréchet in 1993. Several numbers of dendrimers were constructed through chemoselective and orthogonal approaches by the discovery of click chemistry concept. The click concept was introduced in 2001 by K. B. Sharpless and co-workers. They found following observation by click chemistry concept.

- The products yields was very close to 100% and side products was null.
- The isolation of products was easy.

Nowadays, copper-catalyzed Huisgen 1,3-dipolar cycloaddition (CuAAC) between azides and terminal acetylenes, is the most studied click reaction, which selectively forms the 1,4-substituted 1,2,3-triazole (Scheme 1). The exothermic CuAAC reaction displays a large thermodynamic driving force (i.e. 84 kJ·mol<sup>-1</sup>) to favor a single product.

In 2004, the CuAAC reaction was first reported as a chemical reaction for the conventional growth synthesis of triazole containing dendrimers [Meldal (2008)]. Presently, the click concept has proven to be a reliable synthetic platform to deliver complex macrostructures. The concept is today extended to include the Diels–Alder (DA) cycloaddition as well as the thiol–ene coupling (TEC), **Scheme 1**. In 2008, Hawker *et al.* illustrated the power of UV, initiated TEC for the divergent growth of generation four polythioether dendrimer. To date, a significant number of dendrimers has been constructed using click concept.

#### Copper catalysed Huisgen 1,3- dipolar cyclo- addition (CuAAC)



#### Diels Alder cycloaddition



#### Thiol-ene coupling (TEC)



**Scheme 1.** Schematic representation of common click reactions.

The orthogonal strategy now closely connected to the click reactions. This concept (click and the orthogonal strategy) was introduced by Malkoch [61] *et al.* in 2007, for the synthesis of four generation dendrimers, via one bis-MPA based and another Fréchet-type dendron [62]. Nowadays, several higher generation dendrimers are synthesized using orthogonal strategy and click coupling [63-67].

#### One pot synthesis

One-pot synthesis was initiated in the 1990's, but this attempt did not give fruitful result and dendrimers obtained were having high PDI i.e. up to 1.4. The first successful attempt regarding one pot system was made by Rannard and Davis *et al.* [68, 69]. They succeeded into synthesizing up to a third generation carbonate dendrimer. For simplicity, one-pot reactions were divided into two major parts, first the non-tandem reactions (NTRs) and the second one tandem reactions (TRs) [70]. NTRs take

place consecutively, one at a time, while TRs take place independently of one another, and simultaneously also. Several dendrimers were also prepared using these approaches [71].

Besides the advantages of these reactions, it was still difficult to synthesize and isolate a perfect monodispersed dendrimer through one-pot reactions. Since formation of byproducts and uncontrolled perfect stoichiometry of the reagents made this method tedious, therefore these strategies are less utilized for the synthesis of monodisperse dendrimers.

**3. Application:** Among branched molecules, dendritic structures became the most innovative class of polymers and advancing entities [72]. Lot of dendrimers have been designed and synthesized on the basis of applications (biological to material applications). These includes liquid crystalline [73-75], chiral [76], micellar [77,78], amphiphilic [79-80], metalodendrimers [81], dendrimer nanoreactor [82,83], dendrimer nanoparticle [84,86], dendrimer-based nanomedicine [87,88], drug delivery [89-91], gene delivery [92-94], light-harvesting [95,193], organoiron-mediated dendrimer syntheses [96], metallocene dendrimers as electrochrome molecular batteries [97,190], functionalized dendrimers [98], dynamers [99], catalysis [100,101], dendrimers nanotube [102], nanocomposites [103], redox aspects [104,105] olefin metathesis [106], dendrimers in solution [107], photoactive dendrimers [108-110], multivalent neoglycoconjugates [111], biohybrid polymer capsules [112], cleavable dendrimers [113,114], dendritic nanomaterials [115], electrode design [116], solubility enhancers [117], gelators [118,119], light-emitting diodes [120-123], environmental remediation [124], MRI agents [125], biomimetics [126], folded dendrimers [127], nonlinear optics [128], quantum dots [129], molecular recognition [130], tailored dendrimers [131] and branched oligogermanes [132].

**4. Different functionalities:** On a molecular level the dendrimers are made up of small basic core units, branch, and periphery, which can incorporate several functionalities into these units. Thus a lot of dendrimers have been constructed incorporating different functionalities either at core, branches, terminal or mixture of any two and three of these dendritic units. These dendrimers are; heterocyclic dendrimers (triazine [133-134], porphyrin [135], thiophene [136,137], pyridines [138,139], thiadiazole [140], carbazole [141,142], oxidazole [143,144] benzothiozole [145], triazoles [146,147] etc), Heteroatoms containing dendrimers (Si, P, B, Ge, or Bi) [148], Different linkage (amide, anhydride, ester, ether, thioether, sulfonamide etc) containing dendrimers [149-155], homoaromatic, and polynuclear aromatic (naphthalene, anthracene, anullenes, etc.) [156-158] aliphatic dendrimers (only alkyl chain, hydroxyl, thiol and conjugated aliphatic chains) [52,73-75,159], dendrimers with biomacromolecules (glycodendrimers [160, 161], purins and pyrimidins [162,163] containing etc) and fullerene dendrimers [164].

Due to a large number of designs and synthetic approaches have been used to create these impressive

structures. Thus a lot of dendrimers have been constructed, and literature is continuously growing, whose description is impossible through one review. Therefore, we shall mainly focus on the synthesis and applications of sulfone-based dendrimers. We will end with its importance in future developments.

#### *The importance of sulfone/ sulfonamide/sulfonimide based molecules*

The development of sulfone-based dendrimers has been energized and elicited by the engrossing importance of sulfone functionality. Which are as following:

- Ease of synthesis of sulfone derivatives. Specialty, N,N-bis-sulfonylation reactions.
- These are the part of various biological systems [165-168].
- The derivative of such molecules may behave like drugs (long range to small acting) [167-169], which shows antibacterial, antimicrobial, antidiabetic, antiretroviral, anticonvulsant, Cox-2 inhibitor and dermatologically based activities.[170]
- Sulfonyl acid chloride is the most active reagents for mono and bis functionalization of amines functionality among carboxylic acid chlorides or the corresponding active esters [171-174]. Sulfonyl acid chloride derivatives are also inexpensively available.
- In the biodegradable process, these linkages are easily cleavable [175].

In spite of biological importance, nowadays, sulfone derivatives also played a major role in the material applications, i.e., sensing, electronic and nanotechnology to nanoscience [176-187].

#### *Need of sulfone-based dendrimers*

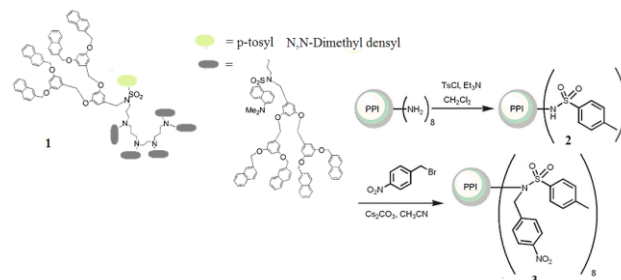
To keep dendrimers monodisperse and to avoid the formation of the side product, which are difficult to separate from the desired dendrimer. There were several methods of synthesis have developed and, continuously been progressing.

Nowadays, construction and functionalization of dendrimers have been successfully achieved through a wide variety of reactions, which are complete, fast, inexpensive and easy to handled. So, sulfone-based moieties come in that category, especially N,N-monosulfonylation and N,N-bissulfonylation based reactions fulfill all the criteria of novel dendrimer synthesis with high efficiency and less cost effectivity. However, the first iteration of N,N-bis-sulfonylation reaction on PAMAM dendrimers was initiated by Vögtle *et al.* [188,189].

#### *Synthesis of sulfone/ sulfonimide /sulfonamide based dendrimers*

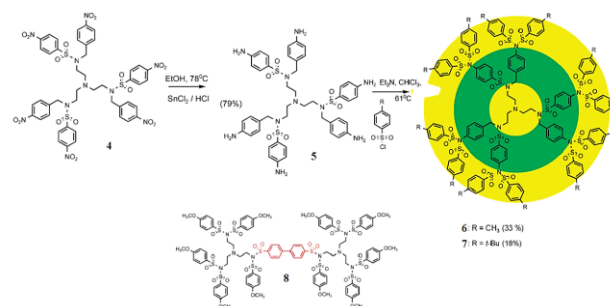
Peripheral functionalization of dendrimers with the activated reagents was the most widely employed method to create dendrimers for possible application. Earlier, in this manner poly(propyleneimine) dendrimers were

selectively functionalized by carboxylic acid chlorides or the corresponding active esters as well as with the isocyanates and isothiocyanates [190-193]. However, these reactions were not generated complete functionalization. Fritz Vögtle *et al.* utilized sulfonic acid chlorides for mono and bifunctionalization of poly(propyleneimine)-dendrimers. They started the N-sulfonylation of poly(propyleneimine)-dendrimers and proposed that the method can be utilized for the synthesis of higher generation sulfonimide based dendrimers [194-196] as shown in Scheme 2.



**Scheme 2.** Structural representation of N-sulfonamide based dendrimers.

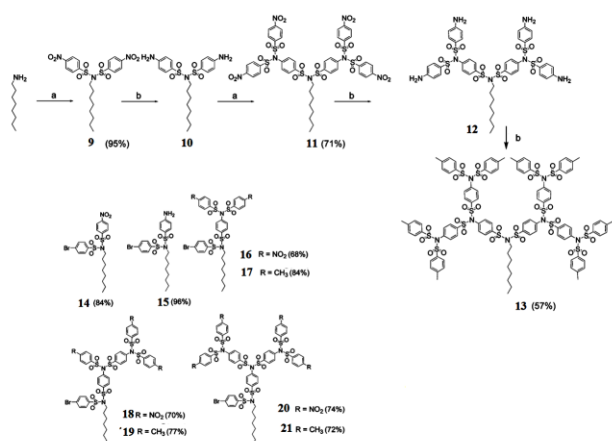
Further, synthesis work of dendrimers, having sulfonimide units at the branching points, were also proceeded by Fritz Vögtle *et al.* in 2004 [197,198]. They gave a method for functionalization of oligoamines and tris(2-aminoethyl) amine via N,N-bis-sulfonylation with various arylsulfonyl chlorides derivatives and standardized reaction conditions utilizing different base. They reduced monosulfonylated precursor 4, using Ethanol/SnCl<sub>2</sub>/HCl at 78 °C, generated precursor 5 and applied it for the preparation of spherical dendrimers 6, and 7 in 33% and 18% yield respectively in divergent fashion (shown in Scheme 3). Similarly, dumbbell-shaped second-generation sulfonamide decorated dendrimer 8 was prepared.



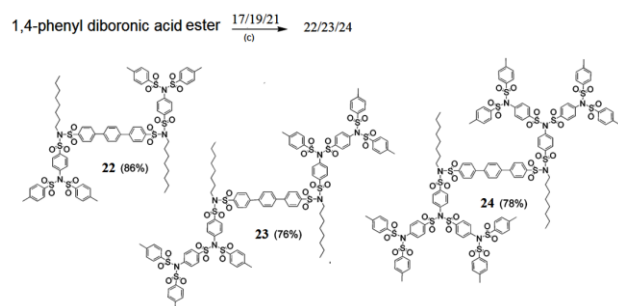
**Scheme 3.** Divergent synthesis of dendrimers possessing N-sulfonimide branching points.

Further, on the basis of mono- or bis-sulfonylation, Oleg Lukin *et al.* constructed several higher generation dendrons and Janus-type dendrimers (up to the third generation) possessing sulfonimide units at every branching point by convergent pathway [199], described in Scheme 4a and 4b. Synthetic steps involved, N-sulfonylation, persulfonylation of amines, reductions of nitroaromatic compounds, and the Suzuki cross coupling (SCC), in which n-octylamine was persulfonylated by p-nitro sulfonyl chloride gave 9 in

95%, which on reduction using HCl/EtOH/SnCl<sub>2</sub> at reflux, led to the formation of generation one (G1) dendron **10**. Further, these steps were repeated and led to the formation of higher generation dendrons **11** and **13** in 71% and 57% respectively. Similarly, dendrons **14-21** were synthesized in decreasing yield as crowding increased (**Scheme 4a**). Further utilization of Suzuki cross-coupling (SCC), dendrimers **22**, **23** and **24** were prepared by the coupling of dendrons **17,19** and **21** to the core 1,4-phenylene diboronic acid diester, in presence of Pd[(P(*p*-To)l)<sub>3</sub>]<sub>3</sub>, THF/H<sub>2</sub>O at 80 °C for 24 h (**Scheme 4b**).



**Scheme 4a:** Synthesis of dendrons and reagent condition (a) ArSO<sub>2</sub>Cl, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>, reflux; (b) SnCl<sub>2</sub>, HCl, EtOH/DCM, at 40 °C.



**Scheme 4b:** Synthesis of dendrimers **22**, **23** and **24** reagent condition (c) Pd[(P(*p*-To)l)<sub>3</sub>]<sub>3</sub>, THF/H<sub>2</sub>O, 80 °C, 24 h.

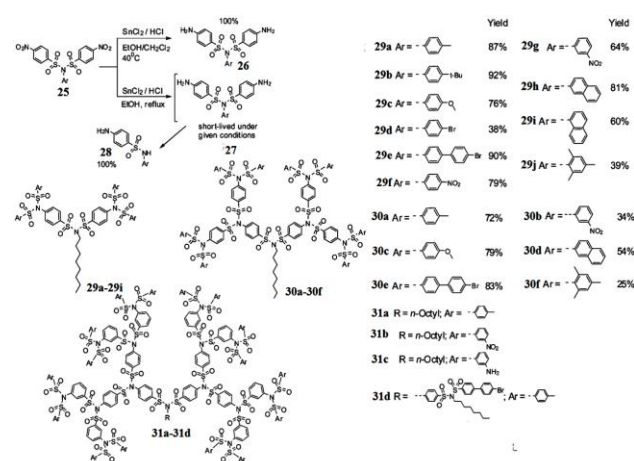
They concluded that -

- Dendrons possessing -NO<sub>2</sub> functionalities (**16**, **18** and **20**) was unable to couple through SCC
- Reduction condition (absolute ethanol and Sn granule) causes cleavage of one of the sulfone linkage.
- Catalytic hydrogenation causes dehydrohalogenation in dendrons having bromoaromatic functionalities.

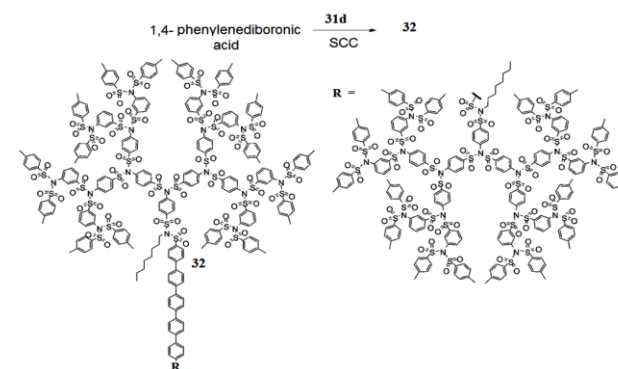
Further in 2007, Oleg Lukin *et al.* overcome the previous problems (reduction condition) [200]. In previous reduction condition, the catalytic hydrogenation caused partial dehydrohalogenation in dendrons having bromoaromatic functionalities (**14-17**). The reduction in refluxing ethanol using SnCl<sub>2</sub> led to the splitproduct **28** through short-lived species **27** (see **Scheme 5a**). On modification of the previous reduction procedure they utilized 1:1 mixture of ethanol/dichloromethane at 40 °C for 20-40 min, in place of absolute ethanol at 78°C, led to

the formation of amine **26** in quantitative yield (100%). Further, combining the new reduction procedure and previously reported monosulfonylation and persulfonylation, several higher generations dendrons were prepared as shown in **Scheme 5a**. In which dendrons **31d** was doubly coupled through core 1,4-phenylenediboronic acid via SCC gave the pentaphenylene centered dendrimer **32** in yield 70% (see **Scheme 5b**).

D. Schubert and Oleg Lukins's group reported a topological study of the above described dendrons and dendrimers [201]. The photophysical study of dendrimer **32** was also started by Vögtle *et al.* [202] and developed by G. Bergamini *et al.*, who gave the photophysical properties (absorption, emission, and excitation spectra; fluorescence decay lifetime and fluorescence anisotropy spectra), using different experimental conditions i.e. dichloromethane solution and solid state at 293 K, dichloromethane/methanol rigid matrix at 77 K [203-206].



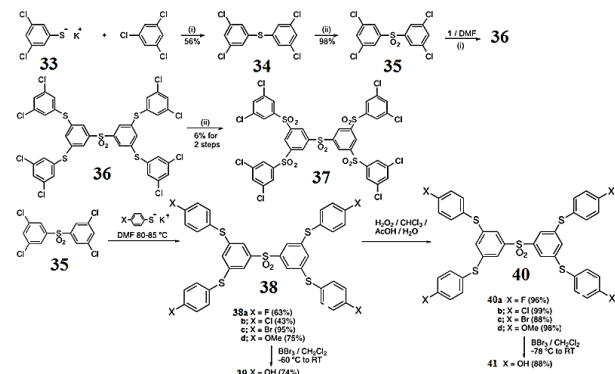
**Scheme 5a:** structural representation of new reduction method and dendrons series.



**Scheme 5b:** Synthesis of dendrimers **32**. Reagent condition of SCC {Pd[(P(*p*-To)l)<sub>3</sub>]<sub>3</sub>, THF/H<sub>2</sub>O, 80 °C, 24 h}.

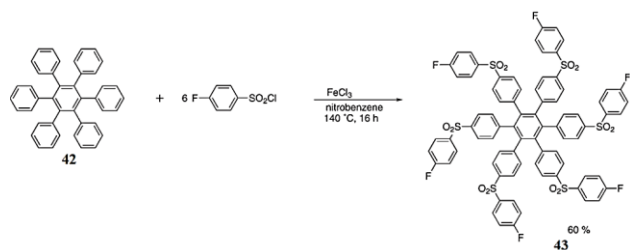
Moreover, Paul C. Taylor developed a divergent route for G1 oligo(aryl sulfone)s and a G2 oligo(aryl sulfone) dendrimer [207, 208], using nucleophilic aromatic substitution reactions, described in **Scheme 6** In this interesting synthesis, they have used compound **33** and trichlorobenzene as starting material and coupled them to prepare **34** which on oxidation of central sulfur gave **35**.

Further, repetition of steps coupling and oxidation, several other intermediate **36**, **37**, **38**, and **40** were constructed. Only **38d** and **40d** were utilized for the construction of dendrimers **39** and **41** in 74% and 80% yield, respectively.



**Scheme 6.** The divergent construction of sulfone-based dendrimers.

In another way of elegant synthesis **Scheme 7** Matsumura *et al.* synthesized novel hexakis(4-(4-fluorophenylsulfonyl)phenyl)benzene **43** in 60% via Friedel Craft reaction. The reaction of hexaphenylbenzene **42** with 4-fluorobenzenesulfonyl chloride in the presence of iron(III) chloride. They utilized this small generation dendrimer for the construction of multiblock copolymers [209].



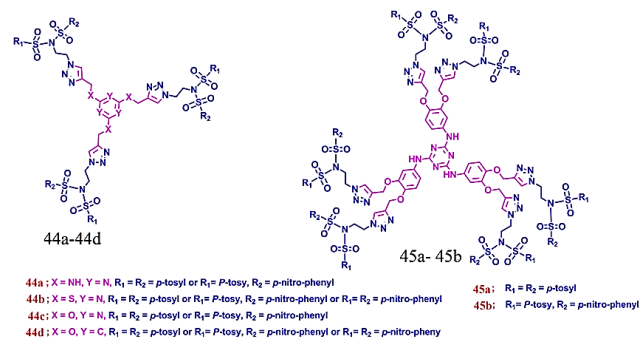
**Scheme 7.** A simple route to sulfone-based dendrimers via FC reaction.

In very recently Tewari *et al.* constructed several triazine cored sulfonimide dendrimers **44a-44 d** (tri-armed) and **45a-45d** (hexa-armed) via click reaction (see **Scheme 8**) [210]. They gave an efficient method of click synthesis and obtained both Cu-free and Cu complexed dendrimers in good yield.

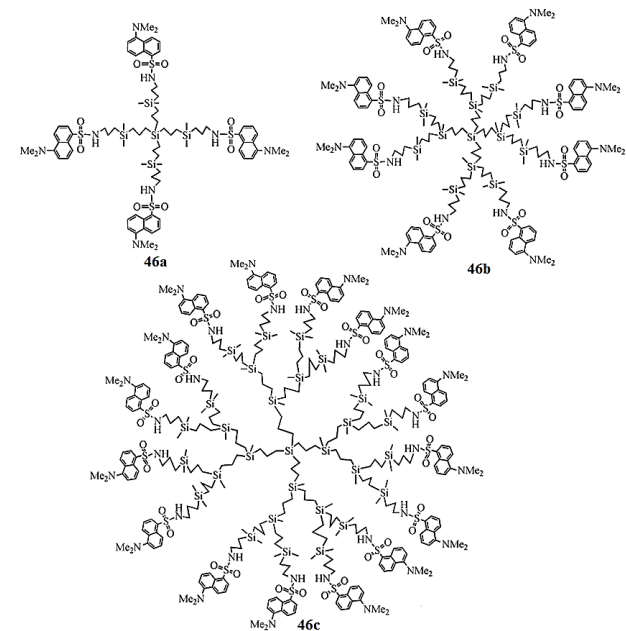
**Application:** Above section described, sulfone/sulfonamide and sulfonimide based, dendrimer synthesis. Nowadays, several sulfone-based dendrimers were synthesized for their important applications. Therefore, in this part, we shall describe synthesis and application of sulfone-based dendrimers

**Biological application:** In the following examples, we shall illustrate work of Paula Ortega *et al.* They utilized the sulfone-based dendrimers (carbosilane dendrimers containing dansyl groups at their periphery as shown in **Scheme 9**) as biologically active tools [211]. Paula Ortega *et al.* developed their previous synthesis of carbosilane dendrimers [212-219], to prove the internalization studies of carbosilane dendrimers. They carried monosulfonylation, at the periphery of

previous dendrimers, by dansylchromophoric group (i.e. 5-(dimethylamino)-1-naphthalenesulfonamido) and prepared dendrimers **46a-46c**. Further, these dendritic systems were tested for *in vitro* toxicity on primary cell cultures (dendritic and peripheral blood mononuclear cells), and their cellular uptake was studied using confocal fluorescence microscopy.



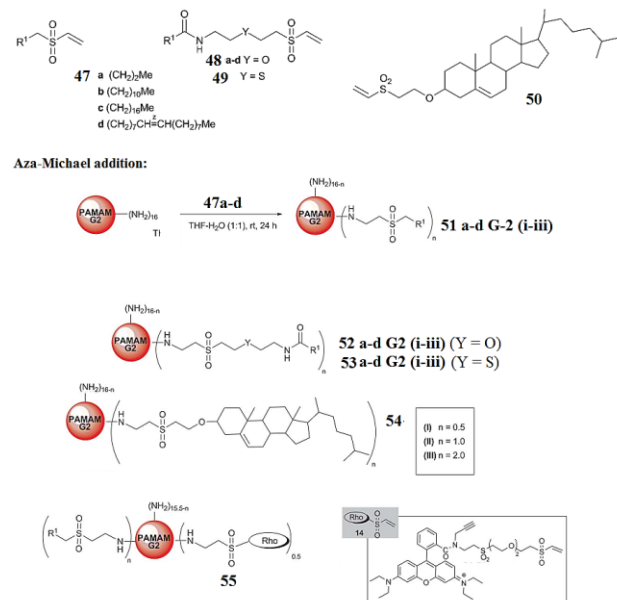
**Scheme 8.** Triazine cored sulfonimide dendrimers **44a-44d** (tri-armed) and **45a-45b** (hexa-armed).



**Scheme 9.** Structural representation of per-dansylated dendrimers **46a-46c**.

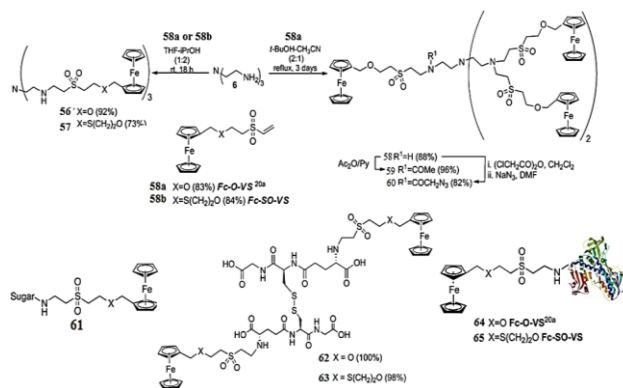
On the other hand, J. M. Sanfrutos utilized PAMAM dendrimers to functionalized their primary amino groups to vinylsulfone group via aza-Michael type addition reaction and prepared upto two generation dendrimer series bearing peripheral alkyl sulfonyl hydrophobic tails (**52-54**) as shown in **Scheme 10** [220]. For this they prepared vinylsulfone series **47**, **48**, **49**, and **50**, which were subsequently treated with the PAMAM amine group in the presence of 1:1 mixture of THF/Water at r.t., gave G1 dendrimers **51a-d**. Similarly, G-2 series **52-55** were prepared. These dendrimers were utilized for gene delivery. For this, they compare the transfection properties of native PAMAM-G2 dendriplexes and amphiphilic sulfone-based dendrimer. They found,

dendrimers (**52-55**) were highly efficient for gene delivery. The incorporation of hydrophobic alkyl chains of different length was anticipated to be a structural factor that would positively contribute to improving the cellular uptake and internalization of the corresponding dendriplexes synergistically. A. M. Fernandez *et al.* created an exciting vinyl sulfone ferrocenylation reagents [182]. The key sequence of reaction is shown in **Scheme 11**. In which compounds **58a** and **58b** were simultaneously coupled to tris (2-aminoethyl) amine (**6**) through Michael-type addition and created G1 dendrimers **56**, **57** and **58**. Azide derivative **60** was synthesized in two steps **i** and **ii** via precursors **58** and **59**. Similarly, recombination of the monomers **58a** and **58b** to the aminosugars, cyclodextrins, tripeptide glutathione (GSH), glutathione disulfide (GSSG) created conjugates and bioconjugates (ferrocenyl terminated dendrimers **57** and ferrocene-sugar **61**, ferrocene-peptide, ferrocene protein conjugates **62** and **63**, ferrocene-cyclodextrin **64** and **65**). These biomolecules were found for strong bioactivity.

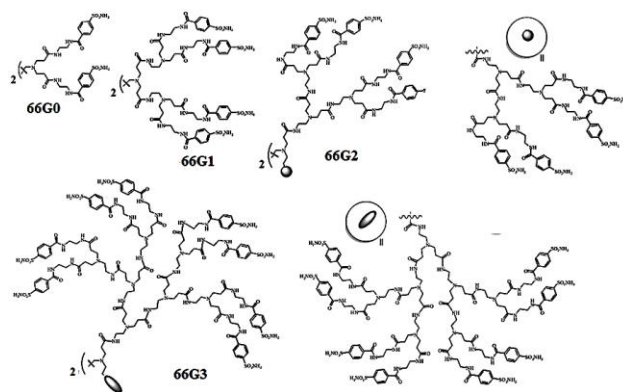


**Scheme 10.** Surface coupling of PAMAM dendrimers via Aza-Michael addition to prepared dendrimers having sulfone units at terminal branches.

In very recent Fabrizio Carta and coworkers also used poly(amidoamine) (PAMAM) dendrimers and their amines were surface functionalized with benzene sulfonyl chloride and created an innovative series of sulfonamide terminated dendrimers **66G0**, **66G1**, **66G2** and **66G3** [222] (see **Scheme 12**). Further, developing their Work, they utilized dendrimers (**66G0-66G3**) for medicinal chemistry applications and investigated benzenesulfonamide moieties of these dendrimers shows strongly human carbonic anhydrase inhibiting properties. Inhibiting properties increases as the sulfonamide molecules increases on going from G0-G3.



**Scheme 11:** Synthesis of bioactive dendrimers possessing sulfone units.

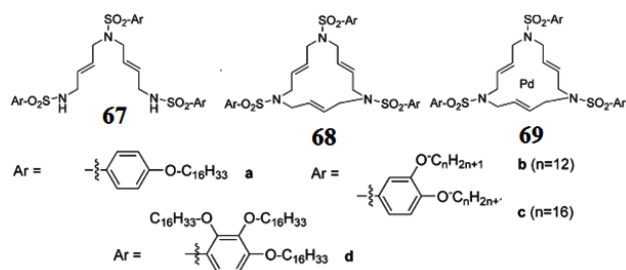


**Scheme 12:** Structure of sulfonamide terminated dendrimers G0-G3.

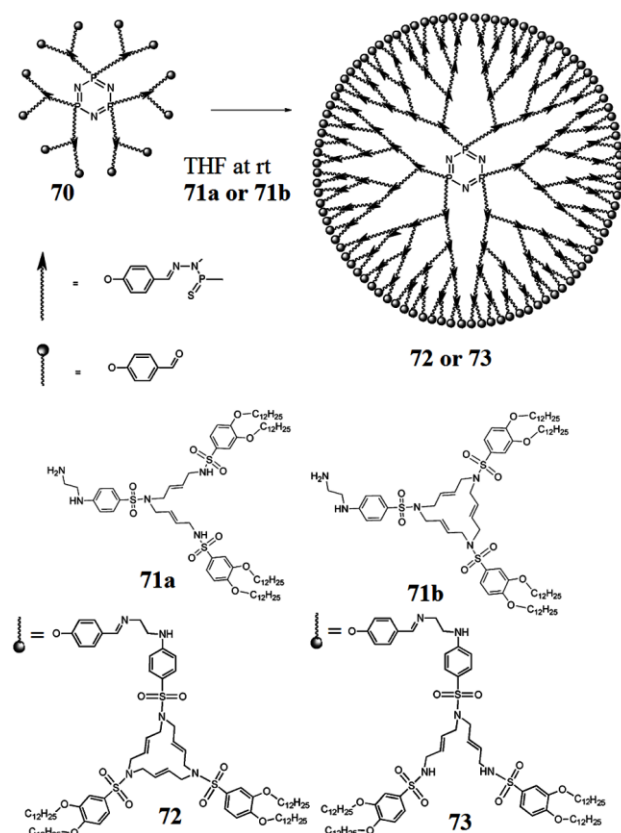
**Material application:** Sulfone based dendrimers constitute one of the most important research topics. In spite of biological applications, some of the sulfone-based dendrimers were constructed, which shows significant interest in materials applications. Now, we shall explore several sulfone-based dendrimers possessing excellent liquid crystalline, non-linear optical and fuel cell properties.

**Liquid crystalline [LC]:** Dendrimers and Dendrons proved particularly versatile candidates for the elaboration of new LC materials [222]. Particularly, the dendritic structure is an interesting framework, where liquid crystalline behavior can be modulated by very subtle modifications of the dendritic connectivity. There have been several kinds of dendrimers constructed such as side-chain liquid crystalline dendrimers, silicon-containing LC dendrimers, polyamide amine (PAMAM), polypropyleneimine (PPI) dendrimers, Polyester, polyether LC dendrimers, ferrocene and/or C-60 subunits containing LC dendrimers. Recently, M. M. Manas and co-workers reported the synthesis of the 15 membered triolefinic aza-macrocycles functionalized with appropriate are sulfonamide groups as well as their palladium(0) complexes **67**, **68** and **69** has been explored in **Scheme 13** [223].





**Scheme 13.** Synthesis of Liquid crystalline dendrimers bearing N-sulfonamide moieties.



**Scheme 14.** Synthesis of N-sulfonamide dendrimers with mesogenic properties.

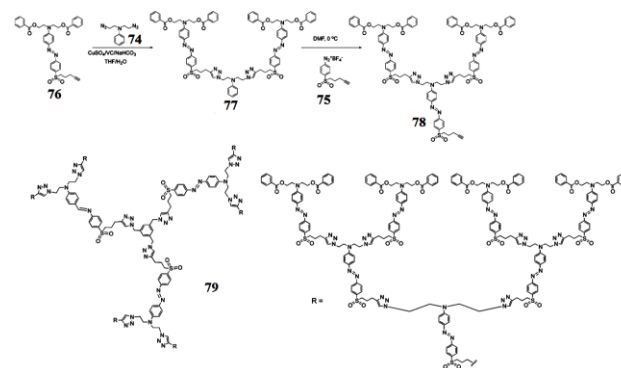
The mesomorphic behavior of the compounds was studied by polarizing optical microscopy, differential scanning calorimetry and X-ray diffraction, which enabled the determination of the structural parameters.

Very recently, C. A. Hincapi et al., by their previous work based on small liquid crystalline molecules, prepared a series of liquid crystalline dendrimers. They constructed liquid crystalline dendrimers by combining phosphorus dendrimers with sulfonamide derivatives. Thus generated dendromesogens **72** and **73** bearing mesogenic units on their surfaces (see **Scheme 14**) [224]. These effects are probably due to the establishment of inter- and intra-molecular hydrogen bonds, generated by the external sulfonamide moieties bearing N-H bonds.

W. Wu and Zen Li's group reported two series of sulfonyl based high-generation NLO dendrimer **79** through the combination of the divergent, convergent

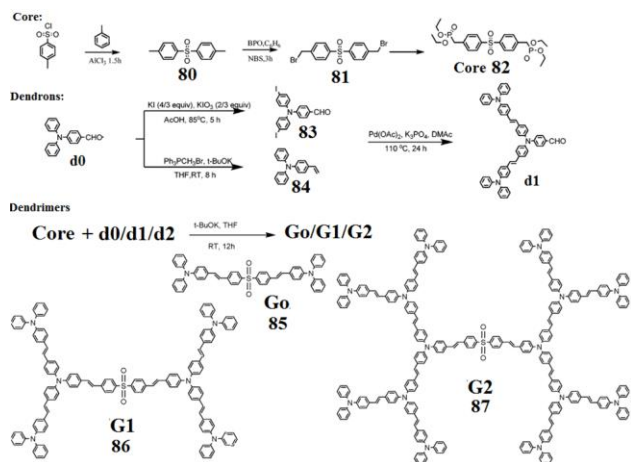
approaches, and Sharpless click reaction, which was easily constructed with high purity and satisfactory yields. Although they previously prepared several nitro groups containing dendrimers, that dendrimers demonstrated high ( $d_{33}$ ) second harmonic generation coefficient values (NLO coefficient). In the present work, they found sulfonyl-based chromophore dendrimers have high optical transparency and NLO stability comparatively without sulfone units. Basic outlook of the synthesis was explained in **Scheme 15**. In which intermediate **76** was clicked to the **74**, led to the formation of **77**, which on diazotization using **75**, gave **78**. Repeating these steps dendritic series have been constructed [225, 226].

Kunpeng Li [227] *et al.* gave an easy convergent synthetic strategy of dendrimer synthesis without protection/deprotection steps. Synthetic steps were a combination of several reactions like Friedel Craft acylation, Vilsmeier reaction, Heck coupling in dendrons and Wittig-Horner reaction, summarized in **Scheme 16**. Donor core **82** and acceptor dendrons **d0**, **d1**, and **d2** were synthesized by applying these reactions, where sulfonylation of toluene gave **80**, which on bromination and phosphorylation led to the formation of **81** and core **82**. Further dendron **d0** was synthesized via Vilsmeier reaction, which acts as a precursor for **83** and **84**. Heck coupling between **83** and **84** in the presence of  $\text{Pd}(\text{oAc})_2/\text{K}_3\text{PO}_4$ , gave dendron **d1**. Similarly, **d2** was synthesized. Further, dendrons **d0-d2** were coupled through Wittig-Horner reaction to the core **82**, led to the formation of novel dendrimers generations (G0-G2) **85**, **86** and **88**. These dendrimers were applied for hetero junction solar cells, in which dendrimer G0 shows a relatively high power conversion efficiency (PCE) of 0.34% under AM 1.5 illumination of  $100 \text{ mWcm}^{-2}$ .

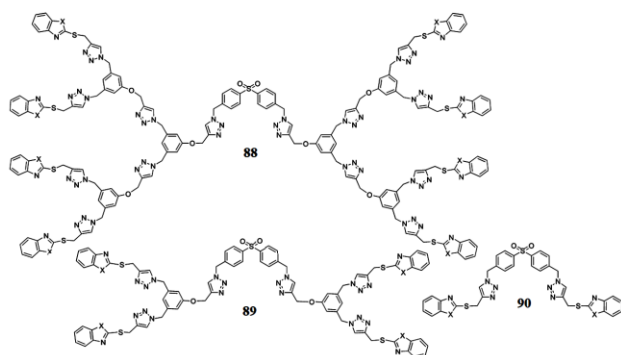


**Scheme 15.** Convergent synthesis of dendrimers via CuAAC having NLO properties.

In **Scheme 17**, the sulfone core was again utilized by P. Rajakumar et al.. They developed their previous work of benzothiazole terminated dendrimers [228], utilizing the click coupling to construct the dendrimers **88**, **89** and **90**. They used, these dendrimers in dye-sensitized solar cells (DSSC) as additives, which exhibited good current generating capacity and a power conversion efficiency in higher-generation than lower generation dendrimers.

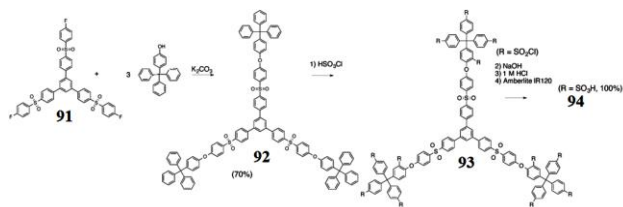


**Scheme 16.** The simple synthesis of sulfone cored dendrimers.



**Scheme 17.** The sulfone cored dendrimers are bearing good DSSC properties.

Recently, M. M. Zerco et al. gave a straightforward procedure for the synthesis of a novel sulfonated dendrimers **93** and **94** with a PES backbone [229]. Dendrimer **94** was blended with poly-[1-(4,4-diphenylether)-5-oxybenzimidazole]benzimidazole (PBIOOVR) at different ratios to prepare a series of acid-base PEMs (**Scheme 18**). They studied these dendrimers as a fuel cell.



**Scheme 18.** Synthesis of the polysulfonated dendrimer **93** and **94** with a PES backbone.

### Future viewpoint

An analysis of the sulfone/sulfonamide/sulfonimide chemical structures, synthesis methods and applications presently available reveals that these dendrimers have not been fully explored yet. Both synthetic and application aspects can be enhanced by introducing heterocyclic, polycyclic hydrocarbon heteroatoms, etc. triazine with sulfone/ sulfonamide or sulfonimide. As a final remark,

combined with advancing functionalities these dendrimers will have a great impact on dendrimer synthesis. It would advance the application of such dendrimers.

### Conclusion

In this article, we deliberated to review the comprehensive description of sulfone-based dendrimers synthesis and their biological to material applications. We have described the type of dendrimers, importance, and need of sulfone derivatives and the chemistry, in short. It was observed that till now several kinds of sulfone-based dendrimers were constructed by utilizing different reactions, but N,N-bis-sulfonylation reaction were fast and complete. Thus several dendritic assemblies were also created that possessing sulfonamide units and have applications. This article is endeavoring to find potential future directions in the development of more analogs of sulfone-based dendrimers synthesis for various biological to material applications. We expected that the information given in this article will not only modernize the researchers with recent synthesis and applications of these dendrimers but also buoy up them to use this promising moiety for the design of new molecules with enriched medicinal and material properties, that are leading to novel treatments luxury agents for the benefit of humanity and technology.

### Acknowledgements

SK, SKR and AKT thanks to Department of Chemistry, Institute of Science of Banaras Hindu University for the facilities.

### References

- Jana, A.; Paikar, A.; Bera, S.; Maity, S.; Haldar, D. *Org. Lett.* **2013**, *16*, 38-41.  
DOI: 10.1021/ol402865t
- Singh, S.; Aggarwal, A.; Bhupathiraju, N. V. S. D. K.; Arianna, G.; Tiwari, K.; Drain, C. M. *Chem. Rev.* **2015**, *115*, 10261–10306.  
DOI: 10.1021/acs.chemrev.5b00244
- Binnemans, K. *Chem. Rev.* **2009**, *109*, 4283-4374.  
DOI: 10.1021/cr8003983
- Singla, P.; Luxami, V.; Paul, K. E. *J. Med. Chem.* **2015**, *102*, 39-57. DOI: 10.1016/j.ejmech
- Aaron, J. L.; Villaraza, A. B.; Martin, W. B. *Chem. Rev.* **2010**, *110*, 2921-2959.  
DOI: 10.1021/cr900232t
- Hoeben, F. J. M.; Jonkheijm, P.; Meijer, E. W.; Schenning, A. P. H. J. *Chem. Rev.* **2005**, *105*, 1491-1546.  
DOI: 10.1021/cr030070z
- Vriezema, D. M.; Aragonés, M. C.; Elemans, J. A. A. W.; Cornelissen, J. J. L. M.; Rowan, A. E.; Nolte, R. J. M. *Chem. Rev.* **2005**, *105*, 1445-1490.  
DOI: 10.1021/cr0300688
- Caminade, A. M.; Ouali, A.; Laurent, R.; Turrin, C. O.; Majoral, J. P. *Coord. Chem. Rev.* **2016**, *308*, 478-497.  
DOI: 10.1016/j.ccr.2015.06.007
- Grimsdale, A. C.; Chan, K. L.; Martin, R. E.; Jokisz, P. G.; Holmes, A. B. *Chem. Rev.* **2009**, *109*, 897-1091.  
DOI: 10.1021/cr000013v
- Grayson, S. M.; Fréchet, J. M. J. *Chem. Rev.* **2001**, *101*, 3819-3868.  
DOI: 10.1021/cr990116h
- Fujita, K.; Hattori, H. *Tetrahedron.* **2016**, *64*, 1485-1492.  
DOI: 10.1016/j.tet.2016.01.048
- Yao, Q.; Kinney, E. P.; Yang, Z. *J. Org. Chem.* **2003**, *68*, 7528-7531.  
DOI: 10.1021/jo034646w


13. Günay, K. A.; Theato, P.; Klok, H. A. *J. Polym. Sci. Part A: Polym. Chem.* **2013**, *51*, 1-28.  
DOI: 10.1002/pola.26333
14. Tanaka, T.; Arimoto, R.; Sakai, S.; Fujii, M.; Yamamoto, M. K. *J. Am. Chem. Soc.* **2005**, *127*, 13896-13905.  
DOI: 10.1021/ja0524797
15. Suzuki, M.; Nakajima, R.; Tsuruta, M.; Higuchi, M.; Einaga, Y.; Yamamoto, K. *Macromolecules*. **2006**, *39*, 64-69.  
DOI: 10.1021/ma052268m
16. Chinchilla, R.; Nájera, C. *Chem. Rev.* **2007**, *107*, 874-922.  
DOI: 10.1039/C4RA09105A
17. Karak, M.; Barbosa, L. C. A.; Hargaden, G. C. *RSC. Adv.* **2014**, *4*, 53442-53466.  
DOI: 10.1039/C4RA09105A
18. Meldal, M.; Tornøe, C. W. *Chem. Rev.* **2008**, *108*, 2952-3015.  
DOI: 10.1021/cr0783479
19. Thirumurugan, P.; Matosiuk, D.; Jozwiak, K. *Chem. Rev.* **2013**, *113*, 4905-5028. DOI: 10.1021/cr3003888
20. Kose, M. M.; Gulen, Y. S. A. *Org. Lett.* **2008**, *10*, 2353-2356.  
DOI: 10.1021/ol800553t
21. Tong, R.; Tang, M. L. L.; Tu, C.; Baumgartner, R. C. *Chem. Soc. Rev.* **2014**, *43*, 6982-7012.  
DOI: 10.1039/c4cs00133h
22. Zheng, Y.; Li, S.; Wenga, Z.; Gao, Chao; *Chem. Soc. Rev.* **2015**, *44*, 4091-4130.  
DOI: 10.1039/c4cs00528g
23. Raffa, P.; Zakarias, D. A.; Picchioni, W. F. Broekhuis, A. A. *Chem. Rev.* **2015**, *115*, 8504-8563.  
DOI: 10.1021/cr500129h
24. Bosman, A. W.; Janssen, H. M.; Meijer, E. W. *Chem. Rev.* **1999**, *99*, 1665-1688.  
DOI: 10.1021/cr970069y
25. Flory, P. J. *J. Am. Chem. Soc.* **1941**, *63*, 3096-3100.  
DOI: 10.1021/ja01856a063
26. Flory, P. J. *J. Am. Chem. Soc.* **1952**, *74*, 2718-2723.  
DOI: 10.1021/ja01131a008
27. Vögtle, F.; Buhleier, E. W.; Wehner, W. *Synthesis*. **1978**, *2*, 155-158.  
DOI: 10.1055/s-1978-24702
28. Denkwalter, R. G.; Kolc, J. F.; Lukasavage, W. J. Macromolecular Highly Branched Homogeneous Compound. *U.S. Pat.* **1983**, *4*, 410, 688.
29. Tomalia, D. A. *Prog. Polym. Sci.* **2005**, *30*, 294-324.  
DOI: 10.1016/j.progpolymsci.2005.01.007
30. Newkome, G. R.; Yao, Z.; Baker, G. R.; Gupta, V. K. *J. Org. Chem.* **1985**, *50*, 2003-2004.  
DOI: 10.1021/jo00211a052
31. Newkome, G. R.; Baker, G. R.; Young, J. K.; Traynham, J. G. *J. Polym. Sci., Part A: Polym. Chem.*, **1993**, *31*, 641-651  
DOI: 10.1002/pola.1993.080310307
32. Brabander-van den, B. E. M. M.; Meijer, E. W. *Angew. Chem. Int. Ed. Engl.* **1993**, *32*, 1308-1311.
33. Hawker, C. J.; Fréchet, J. M. J. *J. Am. Chem. Soc.* **1990**, *112*, 7638-7647.  
DOI: 10.1021/ja00177a027
34. L'abbe, G.; Forier, B.; Dehaen, W. *Chem. Commun.* **1996**, *115*, 2143-2144.  
DOI: 10.1039/CC9960002143
35. Moore, J. S. *Acc. Chem. Res.* **1997**, *30*, 402-413.  
DOI: 10.1021/ar950232g
36. Zeng, F.; Zimmerman, S. C. *Chem. Rev.* **1997**, *97*, 1681-1712.  
DOI: 10.1021/cr9603892
37. Jarayaman, N.; Negopodiev, S. A.; Stoddart, J. F. *Chem. Eur. J.* **1997**, *3*, 1193-1199.  
DOI: 10.1002/chem.19970030804
38. Newkome, G. R.; He, E.; Moorefield, C. N. *Chem. Rev.* **1999**, *99*, 1689-1746. DOI: 10.1021/cr9800659
39. Majoral, J. P.; Caminade, A. M. *Chem. Rev.* **1999**, *99*, 845-880.  
DOI: 10.1021/cr970414
40. Vanherck, K.; Koeckelberghs, G.; Vankelecom, I. F. *Prog. Polym. Sci.* **2013**, *38*, 874-896.  
DOI: 10.1016/j.progpolymsci.2012.11.001
41. Abbasi, E.; Aval, S. F.; Akbarzadeh, A.; Milani, M.; Nasarbadi, H. T.; Joo, S. W.; Hanifehpour, Y.; Kazem, N.-K.; Roghiyeh, P.-S. *Nan. Res. Lett.* **2014**, *9*, 1-10.  
DOI: 10.1016/j.nrl.2014.09.027
42. Caminade, A. M.; Ouali, A.; Laurent, R.; Turrin, C. O.; Majoral, J. P. *Chem. Soc. Rev.* **2015**, *44*, 3890-3899.  
DOI: 10.1039/C4CS00261J
43. Ornelas, C. *Macromol. Chem. Phys.* **2016**, *217*, 149-174.  
DOI: 10.1002/macp.201500393
44. Li, W. S.; Kim, K. S.; Jiang, D. L.; Suna, Y.; Aida, T. *J. Am. Chem. Soc.* **2005**, *127*, 7700-7702.  
DOI: 10.1021/ja0513335
45. Joralemon, J. M.; O'Reilly, R. K.; Matson, J. B.; Nugent, A. K.; Hawker, C. J.; Wooley, K. L. *Macromolecules*. **2005**, *38*, 5436-5443.  
DOI: 10.1021/ma050302r
46. Fang, Z.; Teo, T. L.; Cai, L.; Lai, Y. H.; Samoc, A.; Samoc, M. *Org. Lett.* **2009**, *11*, 1-4.  
DOI: 10.1021/ol801238n
47. Aoi, K.; Itoh, K.; Okada, M. *Macromolecules*. **1997**, *30*, 8072-8074.  
DOI: 10.1021/ma961397n
48. Aoi, K.; Motoda, A.; Ohno, M.; Tsutsumiuchi, K.; Okada, M.; Imae, T. *Polym. J.* **1999**, *31*, 1071-1078.  
DOI: 10.1295/polymj
49. Kozaki, M.; Tujimura, H.; Suzuki, S.; Okada, K. *Tetrah. Lett.* **2008**, *49*, 2931-2934.  
DOI: 10.1016/j.tetlet.2008.03.008
50. Tripathy, S.; Das, M. K. *J. App. Pharm. Sci.* **2013**, *3*, 142-149.  
DOI: 10.7324/JAPS.2013.3924
51. Caminade, A. M.; Laurent, R.; Nicot, B. D.; Majoral, J. P. *New J. Chem.* **2012**, *36*, 217-226.  
DOI: 10.1039/C1NJ20458K
52. Newkome, G. R.; Yao, Z. Q.; Baker, G. R.; Gupta, K. *J. Org. Chem.* **1985**, *50*, 2003-2004.  
DOI: 10.1021/jo00211a052
53. Walter, M. V.; Malkoch, M. *Chem. Soc. Rev.* **2012**, *41*, 4593-4609.  
DOI: 10.1039/C2CS35062A
54. Jee, J.-A.; Spagnuolo, L. A.; Rudick, J. G. *Org. Lett.* **2012**, *13*, 3292-3295.  
DOI: 10.1021/ol301263v
55. Caminade, A. M.; Majoral, J. P. *Molecules*, **2016**, *21*, 538.  
DOI: 10.3390/molecules21040538
56. Morikawa, A. *Molecules*. **2016**, *21*, 219.  
DOI: 10.3390/molecules21020219
57. Abbassi, L.; Chabre, Y. M.; Kottari, N.; Arnold, A. A.; André, S.; Jossierand, J.; Gabiusb, H.-J.; Roy, R. *Polym. Chem.* **2015**, *6*, 7666-7683.
58. Wooley, K. L.; Hawker, C. J.; Fréchet, J. M. J. *J. Am. Chem. Soc.* **1991**, *113*, 4252-4261.  
DOI: 10.1021/ja00011a031
59. Xu, Z. F.; Kahr, M.; Walker, K. L.; Wilkins, C. L.; Moore, J. S. *J. Am. Chem. Soc.* **1994**, *116*, 4537-4550.  
DOI: 10.1021/ja00090a002
60. Kawaguchi, T.; Walker, K. L.; Wilkins, C. L.; Moore, J. S. *J. Am. Chem. Soc.* **1995**, *117*, 2159-2165.  
DOI: 10.1021/ja00113a005
61. Malkoch, M.; Malmstrom, E.; Hult, A. *Macromolecules*, **2002**, *35*, 8307-8314.  
DOI: 10.1021/ma0205360
62. Hawker, C. J.; Chu, F. *Macromolecules*, **1996**, *29*, 4370-4380.  
DOI: 10.1021/ma9516706
63. Barany, G.; Merrifield, R. B. *J. Am. Chem. Soc.* **1977**, *99*, 7363-7365.  
DOI: 10.1021/ja00464a050
64. Spindler, R.; Fréchet, J. M. J. *J. Chem. Soc. Perkin Trans. 1*, **1993**, *8*, 913-918.  
DOI: 10.1039/P19930000913
65. Kempe, K.; Krieg, A.; Becer, C. R.; Schubert, U. S. *Chem. Soc. Rev.* **2012**, *41*, 176-191.  
DOI: 10.1039/c1cs15107j
66. Kottari, N.; Chabre, Y. M.; Shiao, T. C.; Rej, R.; Roy, R. *Chem. Commun.* **2014**, *50*, 1983-1985.  
DOI: 10.1039/c3cc46633g
67. Lowe, A. B. *Polym. Chem.* **2010**, *1*, 17-36.  
DOI: 10.1039/b9py00216b
68. Yamakawa, Y.; Ueda, M.; Takeuchi, K. Asai, M. *J. Polym. Sci. Part A: Polym. Chem.* **1999**, *37*, 3638-3645.

- DOI: 10.1002/(SICI)1099-0518(19990915)37:18<3638::AID-POLA12>3.0.CO;2-F
69. Lundberg, P.; Hawker, C. J.; Hult, A.; Malkoch, M. *Macromol. Rapid Commun.* **2008**, *29*, 998–1015.  
DOI: 10.1002/marc.200800181
70. Rannard, S. P.; Davis, N. J. *J. Am. Chem. Soc.* **2000**, *122*, 11729–11730.  
DOI: 10.1021/ja002613h
71. Malkoch, M.; Schleicher, K.; Drockenmuller, E.; Hawker, C. J.; Russell, T. P.; Wu, P.; Fokin, V. V. *Macromolecules.* **2005**, *38*, 3663–3668.  
DOI: 10.1021/ma047657f
72. Newkome, G. R.; Shreiner, C. D. *Polymer.* **2008**, *49*, 1–173.  
DOI: 10.1016/j.polymer.2007.10.021.
73. Tschierske, C. *Chem. Soc. Rev.* **2007**, *36*, 1930–1970.  
DOI: 10.1039/B615517K
74. Buathong, D. S. W.; Bury, I.; Guillon, I. D. *Chem. Soc. Rev.* **2007**, *36*, 1495–1513.  
DOI: 10.1039/B605531C
75. Marcos, M.; Rapún, R. M.; Omenat, A.; Serrano, J. L. *Chem. Soc. Rev.* **2007**, *36*, 1889–1901.  
DOI: 10.1039/B611123H.
76. Roy, R.; Shiao, T. C. *Chem. Soc. Rev.* **2015**, *44*, 3924–3941.  
DOI: 10.1039/C4CS00359D.
77. Yu, Q.; Liu, J.; Chen, D.; Wang, R. *Polymer.* **2015**, *79*, 179–186.  
DOI: 10.1016/j.polymer.2015.10.026.
78. Xiao, F.-X.; Pagliaro, M.; Xu, Y.-J.; Liu, B. *Chem. Soc. Rev.* **2016**, *45*, 3088–3121.  
DOI: 10.1039/C5CS00781J.
79. Liu, X.; Zhou, J.; Tu, T.; Chen, C.; Cheng, Q.; Sengupta, K.; Haung, Y.; Li, H.; Liu, C.; Wang, Y.; Possocco, P.; Wang, M.; Cui, Q.; Giorgio, S.; Fermeglia, M.; Qu, F.; Pricl, S.; Shi, Y.; Liang, Z.; Rochhi, P.; Rossi, J. J.; Peng, L. *Ang. Chem. Int. Ed. Engl.* **2014**, *53*, 11822–11827.  
DOI: 10.1002/anie.201406764
80. Hu, J.; Zhang, G.; Ge, Z.; Liu, S. *Prog. Pol. Sci.* **2014**, *39*, 1096–1143.  
DOI: 10.1016/j.progpolymsci.2013.10.006.
81. Fan, W.; Sun, B.; Ma, J.; Li, X.; Tan, H.; Xu, L. *Chem. A Eur. J.* **2015**, *21*, 12947–12959.  
DOI: 10.1002/chem.201501282.
82. Hannes P. L.; Su, G. Y.; Shang, M.; Hessel, V.; Luque, R.; Noël, T. *Chem. Soc. Rev.* **2016**, *45*, 83–117.  
DOI: 10.1039/C5CS00447K.
83. Deraedt, C.; Pinaud, N.; Astru, D. *J. Am. Chem. Soc.* **2014**, *136*, 12092–12098.  
DOI: 10.1021/ja5061388.
84. Bahrami, K.; Arabi, M. S. *New J. Chem.*, **2016**, *40*, 3447–3455.  
DOI: 10.1039/C5NJ03219A.
85. Duan, H.; Wang, D.; Li, Y. *Chem. Soc. Rev.* **2015**, *44*, 5778–5792.  
DOI: 10.1039/C4CS00363B.
86. Gawande, M. B.; Goswami, A.; Asefa, T.; Guo, H.; Biradar, A. V.; Peng, D. L.; Zboril, R.; Varma, R. S. *Chem. Soc. Rev.* **2015**, *44*, 7540–7590.  
DOI: 10.1039/C5CS00343A.
87. Caminade, A. M. *Chem. Soc. Rev.* **2016**.  
DOI: 10.1039/c6cs00074f.
88. Chen, G.; Roy, I.; Yang, C.; Prasad, P. N. *Chem. Rev.* **2016**, *116*, 2826–2885.  
DOI: 10.1021/acs.chemrev.5b00148.
89. Medina, S. H.; El-Sayed, M. E. H. *Chem. Rev.* **2009**, *109*, 3141–3157.  
DOI: 10.1021/cr900174j
90. Kesharwani, P.; Jain, K.; Jain, N. K. Dendrimer as nanocarrier for drug delivery. *Prog. Poly. Sci.* **2014**, *39*, 268–307.  
DOI: 10.1016/j.progpolymsci.2013.07.005
91. Parajapati, S. K.; Maurya, S. D.; Das, M. K.; Tilak, V. K.; Dhakar, R. C. *J. Drug Deliv. & Therap.* **2016**, *6*, 67–72.  
DOI: 2250-1177.
92. Karimi, M.; Ghasemi, A.; Zangabad, P. S.; Rahighi, R.; Basri, S. M. M.; Mirshekari, H.; Bozorgomid, M. Z. S.; Ghosh, M.; Beyzavi, D. A.; Vaseghi, A.; Aref, A. R.; Haghani, L.; Bahrami, S.; Hamblin, M. R. *Chem. Soc. Rev.* **2016**, *45*, 1457–1501.  
DOI: 10.1039/C5CS00798D.  
DOI: 10.1039/C2CS35306G.
93. Liu, K.; Xu, Z.; Yin, M. *Prog. Poly. Sci.* **2015**, *46*, 25–54.  
DOI: 10.1016/j.progpolymsci.2014.11.005
94. Dichtel, W. R.; Hecht, S.; Fréchet, J. M. J. *Org. Lett.* **2005**, *20*, 4451–4454.  
DOI: 10.1021/ol0516824.
95. Astruc, D.; Ruiz, J. *Tetrah.* **2010**, *66*, 1769–1785.  
DOI: 10.1016/j.tet.2009.11.116.
96. Astruc, D.; Ornelas, C.; Ruiz, J. *Chem. Eur. J.* **2009**, *15*, 8936–8944.  
DOI: 10.1002/chem.200901294.
97. Ambade, A. V.; Chen, Y.; Thayumanavan, S. *New J. Chem.* **2007**, *31*, 1052–1063.  
DOI: 10.1039/B617628C.
98. Lehn, J. M. *Prog. Polym. Sci.* **2005**, *30*, 814–831.  
DOI: 10.1016/j.progpolymsci.2005.06.002
99. Maity, P.; Yamazoe, S.; Tsukuda, T. *ACS Catal.*, **2013**, *3*, 182–185.  
DOI: 10.1021/cs3007318
100. Blanco, V.; Leigh, D. A.; Marcos, V. *Chem. Soc. Rev.* **2015**, *44*, 5341–5370.  
DOI: 10.1039/C5CS00096C
101. Feng, C. L.; Zhong, X. H.; Steinhart, M. Caminade, A. M.; Majoral, J. P.; Knoll, W. *Small.* **2008**, *4*, 566–571.  
DOI: 10.1002/smll.200700453
102. Tang, Y. H.; Huang, A. Y.; Chen, P. Y.; Chen, H. T.; Kao, C. L. *Curr Pharm Des.* **2011**, *22*, 2308–2330.
103. Taranekar, P.; Fulghum, T.; Patton, D.; Ponnappati, R.; Clyde, G.; Advincula, R. *J. Am. Chem. Soc.* **2007**, *129*, 12537–12548.  
DOI: 10.1021/ja074007t.
104. Bhattacharya, P.; Kaifer, A. E. *J. Org. Chem.* **2008**, *73*, 5693–5698.  
DOI: 10.1021/jo800953v.
105. Astruc, D. *Oil Gas Sci. Technol.* **2007**, *62*, 787–797.  
DOI: org/10.2516/ogst:2007055.
106. Ballauff, M.; Likos, C. N. *Angew. Chem. Int. Ed.* **2004**, *43*, 2998–3020.  
DOI: 10.1002/anie.200300602.
107. Balzani, V.; Ceroni, P.; Juris, A.; Venturi, M.; Campagna, S.; Puntoriero, F.; Serroni, S. *Coord. Chem. Rev.* **2001**, *545*, 219–221.  
DOI: 10.1016/S0010-8545(01)00351-4.
108. Ceroni, P.; Bergamini, G.; Marchioni, F.; Balzani, V. Luminescence as a tool to investigate dendrimer properties. *Prog. Polym. Sci.* **2005**, *30*, 453–473.  
DOI: 10.1016/j.progpolymsci.2005.01.003.
109. Chabre, Y. M.; Roy, R.; Horton, D.; Diego, C. A. S. In *Advances in Carbohydrate Chemistry and Biochemistry* Ed. Elsevier. **2010**, *63*, 165–393.  
DOI: 10.1016/S0065-2318(10)64005-X.
110. Marturano, V.; Cerruti, P.; Carfagna, C.; Giamberini, M.; Tytkowski, B.; Ambrogio, V. *Polymer.* **2015**, *70*, 222–230.  
DOI: 10.1016/j.polymer.2015.05.059.
111. Dongen, V.; De Hoog, S. F. M.; Peters, H. P.; Nallani, R. J. R.; Nolte, M.; Van, R. J. M.; Hest, J. C. M. *Chem. Rev.* **2009**, *109*, 6212–6274.  
DOI: 10.1021/cr900072y.
112. Kevitch, R. M.; McGrath, D. V. *New J. Chem.* **2007**, *31*, 1332–1336.  
DOI: 10.1039/B617289J.
113. Sun, X.; Wang, Y.; Lei, Y. *Chem. Soc. Rev.* **2015**, *44*, 8019–8061.  
DOI: 10.1039/C5CS00496A.
114. Lowe, A. B. *Polymer.* **2014**, *55*, 5517–5549.  
DOI: 10.1016/j.polymer.2014.08.015.
115. Gawande, M. B.; Goswami, A.; Asefa, T.; Guo, H.; Biradar, A. V.; Peng, D.-L.; Zboril, R.; Varma, R. S. *Chem. Soc. Rev.* **2015**, *44*, 7540–7590.  
DOI: 10.1039/C5CS00343A
116. Gupta, U.; Agashe, H. B.; Asthana, A.; Jain, N. K. *Biomacromolecules.* **2006**, *7*, 649–658.  
DOI: 10.1021/bm050802s
117. Sangeetha, N. M.; Maitra, U. *Chem. Soc. Rev.* **2005**, *34*, 821–836.  
DOI: 10.1039/B417081B.
118. Yang, X.; Zhou, G.; Wong, W.-Y. *Chem. Soc. Rev.* **2015**, *44*, 8484–8575.  
DOI: 10.1016/j.polymer.2015.09.073
119. Hwang, S. H.; Moorefield, C. N.; Newkome, G. R. *Chem. Soc. Rev.* **2008**, *37*, 2543–2557.  
DOI: 10.1039/B803932C.
120. Xu, F.; Kim, H. U.; Kim, J.-H.; Jung, B. J.; Grimsdale, A. C.; Hwang, D.-H. *Prog. Polym. Sci.* **2015**, *47*, 92–121.  
DOI: 10.1016/j.progpolymsci.2015.01.005

122. Liu, Y.; Guan, Y.; Zhang, Y. *Polymer*. **2015**, *77*, 366-376.  
DOI: 10.1016/j.polymer.2015.09.073
123. Yang, X.; Zhou, G.; Wong, W.-Y. *Chem. Soc. Rev.* **2015**, *44*, 8484-8575  
DOI: 10.1039/C5CS00424A
124. Zhang, W.; Zhu, S.; Luque, R.; Han, S.; Hu, L.; Xu, G. *Chem. Soc. Rev.* **2016**, *45*, 715-752.  
DOI: 10.1039/C5CS00297D
125. Verwilt, P.; Park, S.; Yoon, B.; Kim, J. S. *Chem. Soc. Rev.* **2015**, *44*, 1791-1806.  
DOI: 10.1039/C4CS00336E
126. Han, H.; Zhang, S.; Wang, Y.; Chen, T.; Jin, Q.; Chen, Y.; Li, Z.; Ji, J. *Polymer*. **2016**, *82*, 255-261.  
DOI: 10.1016/j.polymer.2015.11.022.
127. Lockman, J. W.; Paul, N. M.; Parquette, J. R. *Prog. Polym. Sci.* **2005**, *30*, 423-452.  
DOI: 10.1016/j.progpolymsci.2005.01.012
128. Cho, M. J.; Choi, D. H.; Sullivan, P. A.; Akelaitis, A. J. P.; Dalton, L. R. *Prog. Polym. Sci.* **2008**, *33*, 1013-1058.  
DOI: 10.1016/j.progpolymsci.2008.07.007.
129. Tomczak, N.; Janczewski, D.; Han, M.; Vancso, G. J. *Prog. Polym. Sci.*, **2009**, *34*, 393-430.  
DOI: 10.1016/j.progpolymsci.2008.11.004.
130. Li, D.; Chen, Y.; Liu, Z. *Chem. Soc. Rev.* **2015**, *44*, 8097-8123.  
DOI: 10.1039/C5CS00013K
131. Grinstaff, M. W. *J. Polym. Sci. Part A: Polym. Chem.* **2008**, *46*, 383-389.  
DOI: 10.1002/pola.22525
132. Weinert, C. S. *Dalton Trans.* **2009**, *30*, 1691-1699.  
DOI: 10.1039/B816067H.
133. Simenek, E. E.; Abdou, H.; Lalwani, S.; Lim, J.; Mintzer, M.; Venditto, V. J.; Vittur, B. *Proc. R. Soc. A*. **2010**, *466*, 1445-1460.  
DOI: 10.1098/rspa.2009.0108.
134. Bansal, K. K.; Kakde, D.; Gupta, U.; Jain, N. K. *J. Nanoscience and Nanotechnology*. **2010**, *10*, 8395-8404. ISSN: 1533-4899
135. Li, W. S.; Aida, T. *Chem. Rev.* **2009**, *109*, 6047-6076.  
DOI: 10.1021/cr900186c
136. Wu, T.; Shi, J.; Li, C.; Song, J.; Xu, L.; Wang, H. *Org. Lett.* **2013**, *15*, 354-357.  
DOI: 10.1021/ol303300w
137. Hadad, C.; Martínez, J. C. G.; López, R. J. *J. Org. Chem.* **2012**, *77*, 6223-6230.  
DOI: 10.1021/jo301054w
138. Cumpstey, N.; Bera, R. N.; Burn, P. L.; Samuel, I. D. W. *Macromolecules*. **2005**, *38*, 9564-9570.  
DOI: 10.1021/ma050609q
139. Roland, B. K.; Flora, W. H.; Selby, H. D.; Armstrong, N. R.; Zheng, Z. *J. Am. Chem. Soc.* **2006**, *128*, 6620-6625.  
DOI: 10.1021/ja057548w
140. Rajakumar, P.; Ayyavu, T.; Raja, S.; Ganesan, S.; Maruthamuthu, P. *Tetrah. Lett.* **2012**, *53*, 1139-1143.  
DOI: 10.1016/j.tetlet.2011.12.098.
141. Lee, C. C.; Leung, M. K.; Lee, P. Y.; Chiu, T. L.; Lee, H.; Chun, J.; Liu, C.; Chou, P. T. *Macromolecules*. **2012**, *45*, 751-765.  
DOI: 10.1021/ma202433y.
142. El-Khouly, M. E.; Lee, S. H.; Kay, K. Y.; Fukuzumi, S. *New J. Chem.* **2013**, *37*, 3252-3260.  
DOI: 10.1039/C3NJ00770G.
143. Cha, S. W.; Choi, S. H.; Kim, K.; Jin, J. I. *J. Mater. Chem.* **2003**, *13*, 1900-1904. DOI: 10.1039/B302568C
144. Zhao, Z.; Li, J. H.; Chen, X.; Wang, X.; Lu, P.; Yang, Y. *J. Org. Chem.* **2009**, *74*, 383-395.  
DOI: 10.1021/jo802237c.
145. Wang, J. L.; Zhou, Y.; Li, Y.; Pei, J. *J. Org. Chem.* **2009**, *74*, 7449-7456.  
DOI: 10.1021/jo901539a.
146. Rajakumar, P.; Kalpana, V.; Ganesan, S.; Maruthamuthu, P. *Tetrah. Lett.* **2011**, *52*, 5812-5816.  
DOI: 10.1016/j.tetlet.2011.08.126.
148. Thirumurugan, P.; Matosiuk, D.; Jozwiak, K. *Chem. Rev.* **2013**, *113*, 4905-4979.  
DOI: 10.1021/cr200409f
149. Majoral, J. P.; Caminade, A. M. *Chem. Rev.* **1999**, *99*, 845-880.  
DOI: 10.1021/cr9600971.
150. Haridas, V.; Kashmiri, L.; Yogesh, K. S. *Tetrah. Lett.* **2007**, *48*, 4719-4722.  
DOI: 10.1016/j.tetlet.2007.05.023
151. Vieyres, A.; Lam, T.; Gillet, R.; Franc, G.; Castonguay, A.; Kakkar, A. *Chem. Commun.* **2010**, *46*, 1875-1877.  
DOI: 10.1039/B924888A
152. Freeman, A. W.; Fréchet, J. M. J. *Org. Lett.* **1999**, *1*, 685-687.  
DOI: 10.1021/ol9907724
153. Killops, K. L.; Campos, L. M.; Hawker, C. J. *J. Am. Chem. Soc.* **2008**, *130*, 5062-5064.  
DOI: 10.1021/ja8006325
154. Montanez, M. I.; Campos, L. M.; Antoni, P.; Hed, Y.; Walter, M. V.; Krull, B. T.; Khan, A.; Hult, A.; Hawker, C. J.; Malkoch, M. *Macromolecules*. **2010**, *43*, 6004-6013.  
DOI: 10.1021/ma1009935
155. Morgenroth, F.; Reuther, E.; Mullen, K. *Angew. Chem. Int. Ed. Engl.* **1997**, *36*, 631-634.  
DOI: 10.1002/anie.199706311
156. Bergamini, G.; Ceroni, P.; Maestri, M.; Balzani, V.; Lee, S. K.; Vogtle, F. *Photochem. Photobiol. Sci.* **2004**, *3*, 898-905.  
DOI: 10.1039/B408659G.
157. Morgenroth, F.; Reuther, E.; Müllen, K. *Angew. Chem. Int. Ed.* **1997**, *36*, 631-634.  
DOI: 10.1002/anie.199706311
158. Fang, Z.; Teo, T. L.; Cai, L.; Lai, Y. H.; Samoc, A.; Samoc, M. *Org. Lett.* **2009**, *11*, 1-4.  
DOI: 10.1021/ol801238n.
159. Gray, V.; Dzebo, D.; Lundin, A.; Alborzpour, J.; Abrahamsson, M.; Albinsson, B.; Moth-Poulsen, K. *J. Materials Chem. C*. **2015**, *42*, 11111-11121.  
DOI: 10.1039/C5TC02626A
160. Deb, S. K.; Maddux, T. M.; Yu, L. P. *J. Am. Chem. Soc.*, **1997**, *119*, 9079-9080.  
DOI: 10.1021/ja971790o
161. Kushwaha, D.; Tiwari, V. K. *J. Org. Chem.* **2013**, *78*, 8184-8190. DOI: 10.1021/jo4012392.
162. Appelhans, D.; Maculewicz, B. K.; Janaszewska, A.; Lazniewskab, J.; Voit, B. *Chem. Soc. Rev.* **2015**, *44*, 3968-3996.  
DOI: 10.1039/c4cs00339j
163. Lakshminarayanan, A.; Ravi, V. K.; Tatineni, R.; Rajesh, Y. B. R. D.; Maingi, V.; Vasu, K. S.; Madhusudhan, N.; Maiti, P. K.; Sood, A. K.; Das, S.; Jayaraman, N. *Bioconjugate Chem.* **2013**, *24*, 1612-1623.  
DOI: 10.1021/bc400247w.
164. Qamhieh, K.; Nylander, T.; Black, C. F.; Attard, G. S.; Dias, R. S.; Ainalem, M. L. *Phys. Chem. Chemical Physics*. **2014**, *16*, 13112-13122.  
DOI: 10.1039/C4CP01958J
165. Holler, M.; Nierengarten, J. F. *Aust. J. Chem.* **2009**, *62*, 605-623.  
DOI: 10.1071/CH09163.
166. Blunt, J. W.; Copp, B. R.; Keyzers, R. A.; Munro, M. H. G.; Prinsep, M. R. *Nat. Prod. Rep.* **2016**, *33*, 382-386.  
DOI: 10.1039/b805113p
167. Hager, A.; Vrielink, N.; Hager, D.; Lefranc, J.; Trauner, D.; *Nat. Prod. Rep.* **2016**, *33*, 491-522.  
DOI: 10.1039/c5np00096c
168. Henry, R. J. *Bacteriological reviews*. **1943**, *7*, 344-349. ISSN 2347-9531.
169. Otten, H. **1986**, *17*, 689-690.  
DOI: 10.1093/jac/17.6.689
170. McCormack, P. L. *Drugs*. **2011**, *71*, 2457-2489.  
DOI: 10.2165/11208240-000000000-00000
171. Tóth, É.; Pubanz, D.; Vauthey, S.; Helm, L.; Merbach, A. E. *Chem. Eur. J.* **1996**, *2*, 1607-1615.  
DOI: 10.1002/chem.19960021220
172. Wiener, E. C.; Auteri, F. P.; Chen, J. W.; Brechbiel, M. W.; Gansow, O. A.; Schneider, D. S.; Belford, R. L.; Clarkson, R. B.; Lauterbur, P. C. *J. Am. Chem. Soc.* **1996**, *118*, 7774-7782.  
DOI: 10.1021/ja9536126
173. Valério, C.; Fillaut, J.-L.; Ruiz, J.; Guittard, J.; Blais, J.-C.; Astruc, D. *J. Am. Chem. Soc.* **1997**, *119*, 2588-2589.  
DOI: 10.1021/ja964127t
174. Ambure, P.; Roy, K. Understanding the structural requirements of cyclic sulfone hydroxyethylamines as hBACE1 inhibitors against A $\beta$  plaques in Alzheimer's disease: a predictive QSAR approach. *RSC Adv.*, **2016**, *6*, 28171-28186.  
DOI: 10.1039/C6RA04104C

175. Villemagne, B.; Flipo, M.; Blondiaux, N.; Crauste, C.; Malaquin, S.; Leroux, F.; Piveteau, C.; Villeret, V.; Brodin, P.; Villoutreix, B. O.; Sperandio, O.; Soror, S. H.; Wohlkönig, A.; Wintjens, R.; Deprez, B.; Baulard, A. R.; Willand, N. *J. Med. Chem.* **2014**, *57*, 4876-4888.  
**DOI:** 10.1021/jm500422b
176. Searles, S.; Nukina. *S.Chem. Rev.* **1959**, *6*, 1077-1103.  
**DOI:** 10.1021/cr50030a004.
177. Tabrizian, E.; Amoozadeh, A.; Rahmani, S. *RSC Adv.* **2016**, *6*, 21854-21864.  
**DOI:** 10.1039/C5RA20507G.
178. Li, G.; Zhao, C.; Cui, Y.; Rong, T.; Zhua, C.; Na, H. *RSC Adv.* **2016**, *6*, 23025-23032.  
**DOI:** 10.1039/C6RA02321E
179. Lin, C. X.; Zhuo, Y. Z.; Lai, A. N.; Zhang, Q. J.; Zhu, A. M.; Liu, Q. *L.Rsc. Adv.* **2016**, *6*, 17269-17279.  
**DOI:** 10.1039/C5RA22774G
180. Jiang, H.; Wang, R.; Farhan, S.; Zheng, S. **2015**, *64*, 1794-1800. **DOI:** 10.1002/pi.4982.
181. Syrgiannis, Z.; La Parola, V.; Hadad, C.; Lucio, M.; Vazquez, E.; Giacalone, F.; Prato, M. *Ang.Chem. Int. Ed.*, **2013**, *52*, 6480-6483.  
**DOI:** 10.1002/anie.201207362.
182. Megia-Fernandez, A.; Hernandez-Mateo, F.; Santoyo-Gonzalez, F. *Org. & Biom. Chem.* **2013**, *11*, 2586-2596.  
**DOI:** 10.1039/C3OB27209E.
183. Yoshio, M.; Konishi, R.; Sakamoto, T.; Kato, T. *New J. Chem.* **2013**, *37*, 143-147.  
**DOI:** 10.1039/C2NJ40681K.
184. Osano, K.; Turner, S. *RJournal of Polym. Sci., Part A: Polym.Chem.* **2008**, *46*, 958-969.  
**DOI:** 10.1039/c6py00127k.
185. Adam, M. W.; Takahiro, M.; Kazushi, K. *Polym. Chem.* **2016**, *7*, 2389-2394.  
**DOI:** 10.1039/c6py00127k.
186. Takahiro, M.; Junpei, M.; Soichi, M.; Masahiro, W.; Kenji, M. A. *RSC Adv.* **2015**, *5*, 50082-50086.  
**DOI:** 10.1039/C5RA05698E
187. Xundao, L.; Gao, H.; Chen, X.; Hu, Y.; Li, H.; Zhang, *Polym. Chem.* **2016**, *7*, 2904-2912.  
**DOI:** 10.1039/C6PY00036C.
188. Jingmei, X., Zhe, W.; Huixuan, Z.; Hongzhe, N.; Xueyan, L.; Bingxin, L.; *RSC Adv.* **2016**, *6*, 27725-27737.  
**DOI:** 10.1039/C5RA24894A.
189. Quanbing, M.; Yuan, M.; Xin, J.; Xinyuan, Z. *Macro. Chem. Phys.* **2016**, *Mol. Syst. Des. Eng.*  
**DOI:** 10.1039/c5me00015g.
190. Ahmed, M.; Rahman, A.; Schweizer, B. W.; Lukin, O.; Zhang, A.; Schlüter, A. D. *Macromol. Chem. Phys.* **2010**, *211*, 1538-1549.  
**DOI:** 10.1002/macp.201000031
191. Gindt, B. P. Abebe, D. G.; Tang, Z. J.; Lindsey, M. B.; Chen, J.; Elgammal, R. A.; Zawodzinski, T. A.; Fujiwara, T. *J. Mater. Chem. A* **2016**, *4*, 4288-4295.  
**DOI:** 10.1039/C6TA00698A.
192. Astruc, D.; Boisselier, E.; Ornelas, C. *Chem. Rev.* **2010**, *110*, 1857-1959.  
**DOI:** 10.1021/cr900327d.
193. Archut, A.; Gestermann, S.; Hesse, R.; Kauffmann, C.; Vogtle, F. *Synlett.* **1998**, *5*, 546-548.  
**DOI:** 10.1016/j.polymer.2007.10.021
194. Hahn, U.; Gorka, M.; Vogtle, F.; Vicinelli, V.; Ceroni, P.; Maestri, M.; Balzani, V. *Angew. Chem. Int. Ed.* **2002**, *41*, 3595-3598.  
**DOI:** 10.1002/1521-3773(20021004)41:19<3595::AID-ANIE3595>3.0.CO;2-B.
195. Felder, T.; Schalley, C. A.; Fakhrebavi, H.; Lukin, O. A. *Chem. Eur. J.* **2005**, *11*, 5625-5636  
**DOI:** 10.1002/chem.200401236.
196. Meikelburger, H. B.; Rissanen, K.; Vögtle, F. *Chem Ber.*, **1993**, *126*, 1161-1169.  
**DOI:** 10.1002/cber.19931260516.
197. Meikelburger, H. B.; Vögtle, F. *Supramol Chem.* **1993**, *1*, 187-189. **DOI:** 10.1080/10610279308035159.
198. Moors R.; Vögtle, F. Cascade molecules: building blocks, multiple functionalization, complexing units, photoswitches. In: Newkome GR, editor. *Advances in dendritic macromolecules*. Greenwich, Conn.: JAI **1995**, 41-71.
199. Vögtle, F.; Fakhrebavi, H.; Lukin, O. *Org. Lett.* **2004**, *6*, 1075-1078. **DOI:** 10.1021/ol030130n
200. Vogtle, F.; Fakhrebavi, H.; Lukin, O.; Muller, S.; Friedhofen, J. *Eur. J. Org. Chem.* **2004**, 4717-4724.  
**DOI:** 10.1002/ejoc.200400472
201. Lukin, O.; Gramlich, V.; Kandre, R.; Zhun, I.; Felder, T.; Schalley, C. A. *J. Am. Chem. Soc.* **2006**, *128*, 8964-8974.  
**DOI:** 10.1021/ja061606b
202. Lukin, O.; Schubert, D.; Müller, C.; Corda, M.; Kandre, R. *J. Org. Chem.*, **2008**, *73*, 3562-3565.  
**DOI:** 10.1021/jo800179m.
203. Schubert, D.; Corda, M.; Lukin, O.; Brusilowskij, B.; Fiskin, E.; Schalley, C. A. *Eur. J. Org. Chem.* **2008**, *24*, 4148-4156.  
**DOI:** 10.1002/ejoc.200800408.
204. Vögtle, F.; Gorka, M.; Hesse, R.; Ceroni, P.; Maestri, M.; Balzani, V. *Photochem. Photobiol. Sci.*, **2002**, *1*, 45-51.  
**DOI:** 10.1039/B106813J
205. Vincinelli, V.; Ceroni, P.; Maestri, M.; Lazzari, M.; Balzani, V.; Lee, S. K. *Org. Biomol. Chem.* **2004**, *2*, 2207-2213.  
**DOI:** 10.1039/b404463k.
206. Bergamini, G.; Ceroni, P.; Balzani, V.; Villavieja, M. D. M.; Kandre, R.; Zhun, I. *Chem. Phys. Chem* **2006**, *7*, 1980-1984.  
**DOI:** 10.1002/cphc.200600259.
207. Bergamini, G.; Ceroni, P.; Balzani, V.; Kandre, R.; Lukin, O. *Chem. Phys. Chem.* **2009**, *10*, 265-269.  
**DOI:** 10.1002/cphc.200800597
208. Taylor, P. C.; Wall, M. D.; Woodward, P. R. *Tetrah.* **2005**, *61*, 12314-22.  
**DOI:** 10.1016/j.tet.2005.09.103
209. Percec, V.; Barboiu, B.; Grigoras, C.; Bera, T. K. *J. Am. Chem. Soc.* **2003**, *125*, 6503-6516.  
**DOI:** 10.1021/ja034746j.
210. Matsumura, S.; Hlil, A. R.; Hay, A. S. *J. Polym. Sci. Part A Polym. Chem.* **2008**, *46*, 6365-6375.  
**DOI:** 10.1002/pola.22965.
211. Khanam, S.; Rai, S. K.; Verma, D.; Khanna, R. S.; Tewari, A. **2016**, *6*, 56952-56962.  
**DOI:** 10.1039/C6RA09929G
212. Ortega, P.; Serram, M.-J.; Samaniego, R.; dela Mata, F. J.; Gomez, R. A. M.; Fernandez, M. *Org. Biomol. Chem.* **2009**, *7*, 3079-3085. **DOI:** 10.1039/B900942F.
213. Pathak, A. K.; Pathaka, V.; Seitz, L.; Gurcha, S. S.; Besra, G. S.; Riordan, J. M.; Reynolds, R. C. *Bioorg. Med. Chem.* **2007**, *15*, 5629-5650.  
**DOI:** 10.1016/j.bmc.2007.04.012
214. Amoresano, A.; Chiappetta, G.; Pucci, P.; D'Ischia, P. M.; Marino, *G. Anal. Chem.* **2007**, *79*, 2109-2108.  
**DOI:** 10.1021/ac0620361.
215. Hoffmann, K.; Genger, U. R.; Mix, R.; Friedrich, J. F. *J. Fluoresc.* **2006**, *16*, 441-448.  
**DOI:** 10.1007/s10895-006-0076-x.
216. O'Connor, N. A.; Sakata, S. T.; Huide, Z.; Shea, K. *J. Org. Lett.*, **2006**, *8*, 1581-1584.  
**DOI:** 10.1021/ol060130y
217. Mure, M.; Kurtis, C. R.; Brown, D. R.; Rogers, M. S.; Tambyrajah, W. S.; Saysell, C.; Wilmot, C. M.; Phillips, S. E. V.; Knowles, P. F.; Dooley, D. M.; McPherson, M. J. *Biochemistry*; **2005**, *44*, 15083-1594.  
**DOI:** 10.1021/bi0479860.
218. Misra, A.; Mishra, S.; Misra, K. *Bioconjugate Chem.* **2004**, *15*, 638-646.  
**DOI:** 10.1021/bc049976h.
219. Metivier, R.; Leray, I.; Valeur, B. A. *Chem. Commun.* **2003**, *8*, 996-997.  
**DOI:** 10.1039/B301323E.
220. Ceroni, P.; Laghi, I.; Maestri, M.; Balzani, V.; Gestermann, S.; Gorka, M.; Vogtle, F. *New J. Chem.*, **2002**, *26*, 66-73.  
**DOI:** 10.1039/B105497J.
221. Sanfrutos, J. M.; Fernandez, A. M.; Mateo, F. H.; Gonzalez, M. D. G.; Gonzalez, R. S.; Gonzalez, F. S. *Org. Biomol. Chem.*, **2011**, *9*, 851-864.  
**DOI:** 10.1039/C0OB00355G
222. Carta, F.; Osman, S. M.; Vullo, D.; Allothman, Z.; Supuran, C. T. *Org. Biomol. Chem.*, **2015**, *13*, 6453-6457.  
**DOI:** 10.1039/c5ob00715a.

223. Mañas, M. M.; Pleixats, R.; Sebastián, R. M.; Vallribera, A.; Roglans, A. *J. Organomet. Chem.*, **2004**, *689*, 3669-3684.  
**DOI:** 10.1016/j.jorganchem.2004.05.024.
224. Mañas, M. M.; Reichardt, C.; Sebastián, R. M.; Barberá, J.; Serranob, J. L.; Sierra, T. *J. Mater. Chem.* **2005**, *15*, 2210-2219.  
**DOI:** 10.1039/b502561c.
225. Hincapi, C. A.; Sebastián, R. M.; Barber, J.; Serrano, J. L.; Sierra, T.; Majoral, J. P.; Caminade, A. M. *Chem. Eur. J.* **2014**, *20*, 17047-17058.  
**DOI:** 10.1002/chem.201402539.
226. Wu, W.; Xu, G.; Li, C.; Yu, G.; Liu, Y.; Ye, C.; Qin, J.; Li, Z. *Chem. Eur. J.* **2013**, *19*, 6874-6888.  
**DOI:** 10.1002/chem.201203567.
227. Sullivan, P. A.; Rommel, H.; Liao, Y.; Olbricht, B. C.; Akelaitis, A. J. P.; Firestone, K. A.; Kang, J. W.; Luo, J.; Davies, J. A.; Choi, D. H.; Eichinger, B. E.; Reid, P. J.; Chen, A.; Jen, A. K. Y.; Robinson, B. H.; Dalton, L. R. *J. Am. Chem. Soc.* **2007**, *129*, 7523-7530.  
**DOI:** 10.1021/ja068322b
228. Li, K.; Qu, J.; Xu, B.; Zhou, Y.; Liu, L.; Peng, P.; Tian, W. *New J. Chem.*, **2009**, *33*, 2120-2127.  
**DOI:** 10.1039/b9nj00236g.
229. Rajakumar, P.; Kalpana, V.; Ganesanb, S.; Maruthamuthu, P. *New J. Chem.* **2013**, *37*, 3692-3700.  
**DOI:** 10.1039/c3nj00887h.
230. Zarco, M. M.; Titvinidze, G.; Martinez, J. C. G.; Lopez, J. R. *J. Polym Sci Part A Polym Chem.* **2016**, *54*, 69-80.  
**DOI:** 10.1002/pola.27777



**A Monthly Journal**

**Publish your article in this journal**

Advanced Materials Letters is an official international journal of International Association of Advanced Materials (IAAM, [www.iaamonline.org](http://www.iaamonline.org)) published monthly by VBRI Press AB from Sweden. The journal is intended to provide high-quality peer-review articles in the fascinating field of materials science and technology particularly in the area of structure, synthesis and processing, characterisation, advanced-state properties and applications of materials. All published articles are indexed in various databases and are available download for free. The manuscript management system is completely electronic and has fast and fair peer-review process. The journal includes review article, research article, notes, letter to editor and short communications.

Copyright © 2017 VBRI Press AB, Sweden

[www.vbripress.com/aml](http://www.vbripress.com/aml)