

Post-deposition annealing controlled structural and optical properties of RF magnetron sputtered MoO₃ films

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ABSTRACT

MoO₃ films were deposited on Corning glass and silicon substrates held at room temperature (303 K) by RF magnetron sputtering of metallic molybdenum target at a fixed oxygen partial pressure of 4×10^{-4} mbar and sputter pressure of 4×10^{-2} mbar. The as deposited films were annealed in air at different temperatures in the range 473 – 673 K. X-ray diffraction studies suggest that the as-deposited and the films annealed at 473 K were amorphous in nature, while those annealed at 573 and 673 K were polycrystalline with mixed phases of α - orthorhombic and β - monoclinic MoO₃. Scanning electron microscope images of the films annealed at 573 and 673 K exhibited nanoflower like and nanodisk like structures due to improvement in the crystallinity. Fourier transform infrared studies showed the characteristic vibrations of MoO₃ with shift in the vibrational modes of Mo = O and Mo – O – Mo with increase of annealing temperature. The optical absorption edge of the films shifted towards lower wavelengths side with increase of annealing temperature. Optical band gap of as-deposited films was 2.98 eV with refractive index 2.01, while those annealed at 673 K showed the optical band gap of 3.15 eV and refractive index of 2.08. The MoO₃ films annealed at 673 K were of nanocrystalline with crystallite size of 39 nm with optical band gap of 3.15 eV and refractive index of 2.08 were favorable for electron blocking and hole-selective layers in bulk-heterojunction solar cells. Copyright © 2013 VBRI press.

Keywords: RF sputtering; molybdenum oxide; post-annealing; structural; optical properties.



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The lithium ions inserted into α - MoO₃ films lead to an increase in the charge storage capacity [6]. Orthorhombic α - MoO₃ nanoflakes were find applications in the cold cathode based electronics [7]. Recently, nanocrystalline oxide films attracted much for their promising applications in building of nanodevices and nanosensors [8-11]. Kocak et al. [12] realized the catalytic activity of electrodeposited MoO₃ for nitrite oxidation. Allogho and Ashrit [13] reported on the photochromic activity of MoO₃ films prepared by thermal evaporation to gauge the water content. Molybdenum oxide is a promising candidate for a back contact layer for cadmium telluride solar cells in superstrate configuration because of its high work function, possibly reducing the back contact barrier [14]. Various thin film deposition techniques such as thermal evaporation [15], pulsed laser deposition [16, 17], sputtering [5, 18-20], atomic layer deposition [21], electrodeposition [22, 23], spray pyrolysis [2], sol – gel process [6, 24, 25] and hot wire chemical vapour deposition [26] were employed for growth of MoO₃ films. The phase, structure and surface morphology of the deposited films depend critically on the growth parameters and the adopted thin films deposition technique. The post-deposition annealing in air at different

Introduction

Molybdenum oxide (MoO₃) is a material finds potential applications due to its abundance and attractive properties such as catalyst, gas sensing and electrochromism [1-4]. MoO₃ thin films exist in three phases namely: orthorhombic α - phase, monoclinic β - phase and hexagonal phase. The crystal structure of β - MoO₃ in which MoO₆⁺ octahedral form a three dimensional network by sharing the corners with each other and well suited as a catalyst for number of chemical reactions. The amorphous MoO₃ films formed by sol-gel dip coating exhibited electrochromic properties [5].

temperatures leads to the structural modification of the films [3, 5, 9]. Chang et al. [3] reported on the effect of air annealing at temperatures in the range 493 - 823 K in RF magnetron sputtered MoO₃ films. The films annealed between 623 - 673 K were composed mixed phase of α - and β - phase MoO₃, while < 623 K were single phase of β - MoO₃ which is potential for ammonia sensing. Dhanasankar et al. [5] reported on the sol-gel coated MoO₃ films that the optical band gap increased from 3.53 to 3.73 eV with increase of annealing in air at temperatures from 473 to 523 K and at higher temperatures of 723 K it decreased to 3.63 eV. Lin et al. [9] reported on the electron beam evaporated MoO₃ films, and subsequently annealed in N₂ and O₂ for 10 min by rapid thermal annealing. The predominant β - MoO₃ was achieved at annealing temperatures < 573 K, where as α - MoO₃ realized at 673 K. The optical band gap of the films decreased from 3.93 to 3.89 eV with increase of annealing in N₂ from 473 to 773 K. The films annealed in O₂ the band gap increased from 3.93 to 3.97 eV with increase of annealing temperatures from 473 to 773 K. From these studies, it revealed that the crystallographic structure and optical band gap of the MoO₃ films critically depend on the annealing temperatures and ambient of annealing. Uthanna et al [19] reported on the DC magnetron sputtered MoO₃ films formed at room temperature were of amorphous while those deposited at 473 K were orthorhombic phase. The optical band gap increased from 3.03 to 3.28 eV and the refractive index increased from 2.03 to 2.08 with increasing substrate temperatures from 303 to 473 K due to improvement in the crystallinity of the films. In the present investigation, an attempt is made in the deposition of MoO₃ films at room temperature by RF magnetron sputtering and studied the effect of post-deposition annealing in air on the structural and morphological and optical properties.

Experimental

Preparation of MoO₃ thin films

MoO₃ thin films were deposited onto glass and silicon substrates held at room temperature (303 K) by sputtering of pure metallic molybdenum target in oxygen and argon gas mixture using reactive RF magnetron sputtering technique. Metallic molybdenum target (99.99 % pure) with 50 mm diameter and 3 mm thickness was used for sputtering. The sputter chamber was evacuated by employing diffusion pump and rotary pump combination to achieve base pressure of 4×10^{-6} Pa. Pressure in the sputter chamber was measured with digital Pirani – Penning gauge combinations. Oxygen and argon of 99.99 % purity were used as reactive and sputter gases for deposition of films. After achieving ultimate pressure, oxygen gas was introduced to the sputter chamber to a pressure of 4×10^{-4} mbar. Later, argon gas was admitted to the sputter pressure of 4×10^{-2} mbar. The required quantities of oxygen and argon gases were admitted into the sputter chamber through fine controlled needle valves followed by Aalborg mass flow controllers (Model GFC 17). The 150 W power was supplied to the sputter target using RF power generator (Advance Energy Model ATX-600W). The sputter parameters fixed during the growth of the films are given in

Table 1. The as-deposited films were annealed in air for 2 hours at different temperatures in the range 473 – 673 K.

Characterization

The as-deposited and the MoO₃ films annealed at different temperatures were characterized by studying their chemical composition, crystallographic structure and surface morphology, chemical binding configuration and optical properties. The thickness of the deposited films was measured with a mechanical Veeco Dektak (Model 150) depth profilometer. A force of 10 μ N was applied over the sample surface with a swept length of 3 mm for 100 sec. The chemical composition of the films was analysed with energy dispersive X-ray Analysis (Oxford instruments Inca Penta FETx3) attached to a scanning electron microscope (Carl Zeiss, model EVO MA15). The crystallographic structure of the films was determined by using X-ray diffractometer (Bruker D8 Advance Diffractometer) with copper K α radiation wavelength of $\lambda = 0.15406$ nm. The XRD profiles were taken for a 2θ range between 10⁰ and 60⁰ at the steps of 0.05⁰. The surface morphology of the films was analysed with scanning electron microscope (Hitachi SEM Model S-400). The chemical binding configuration of the films formed on silicon substrates was analyzed with Fourier transform infrared spectrophotometer (Nicolet Magana IR 750) was used to record the transmittance in the wavenumber range 300 – 1500 cm⁻¹. The optical transmittance of the films formed on glass substrates was recorded using UV-Vis-NIR double beam spectrophotometer (Perkin - Elmer Spectrophotometer Lambda 950) in the wavelength range 300 - 1500 nm.

Table 1. Sputter parameters fixed during the growth of the MoO₃ films.

Sputter target	: Molybdenum (50 mm \times 3 mm)
Target to substrate distance	: 65 mm
Ultimate pressure	: 4×10^{-6} mbar
Oxygen partial pressure (pO ₂)	: 4×10^{-4} mbar
Sputtering pressure	: 4×10^{-2} mbar
Sputter power	: 150 W
Deposition time	: 120 min

Results and discussion

Composition analysis

The thickness of the deposited films was determined by using Veeco Dektak depth profilometer was in the range 910 - 980 nm. The chemical composition of the MoO₃ films was determined by using energy dispersive X-ray analysis (EDAX). **Fig. 1** shows the EDAX spectrum of the as-deposited MoO₃ film. The spectrum contained characteristic constituents of molybdenum and oxygen in

the film. The chemical composition of the films was determined from the peaks of molybdenum and oxygen. The atomic ratio of oxygen to molybdenum was 2.98: 1. It indicated that the as-deposited films were nearly stoichiometric. It remains almost constant in the films annealed in air at different temperature.

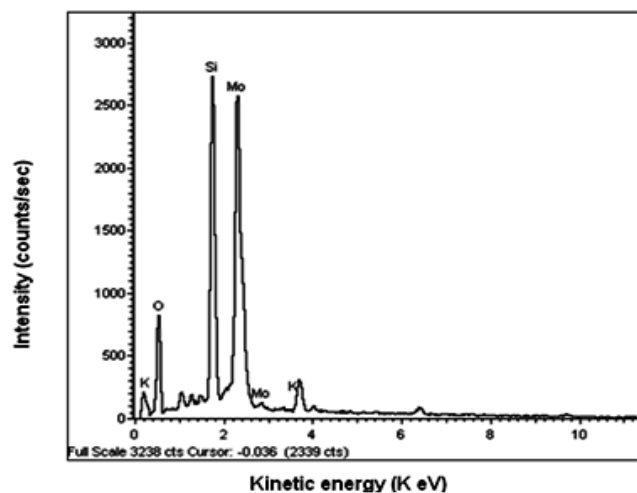


Fig. 1. EDAX spectrum of as deposited MoO₃ film.

XRD study

Fig. 2 shows the XRD profiles of as-deposited and the MoO₃ films annealed at different temperatures. The as-deposited MoO₃ films showed no characteristic diffraction peaks which indicated that the grown films were amorphous in nature. The mobility of ad-atoms arriving on the substrate surface will be low in the as-deposited films and the interaction between the molybdenum and oxygen will be low hence of amorphous nature. The films annealed in air at 473 K were also amorphous (not shown in Fig. 2). It indicated that the thermal energy gained by annealing at 473 K was not sufficient to initiate crystallization of the amorphous phase MoO₃ films. Post-deposition annealing of the films at 573 K exhibited the orientations (020), (040) and (060) related to the growth of orthorhombic α - phase MoO₃ [JCPDS card No. 76 - 1003]. The additional reflections of (011) and (201) were also present, which related to the growth of monoclinic β - phase [JCPDS card No 86 - 1269]. It revealed that the films annealed at 573 K were of mixed phase of orthorhombic α -phase and monoclinic β - phase MoO₃. Further increase of annealing temperature to 673 K, three additional (102), (110) and (400) reflections were also observed due to the formation of β - phase MoO₃ associated with α - phase MoO₃. Thus, the as-deposited and the films annealed at 473 K were amorphous and the crystallinity of films increased with the increase of annealing temperatures ≥ 573 K and associated with the mixed phase of α - phase and β - phase MoO₃. The results achieved are in accordance with the reports of Chang et al. [3] where α - and β - phase films were achieved by annealing of RF magnetron sputtered MoO₃ films in the temperatures in the range 623 - 673 K, and single phase α -MoO₃ films at 723 K. Navas et al. [21] achieved monoclinic Mo₈O₂₃ and orthorhombic MoO₃ phases by annealing of RF magnetron sputtered molybdenum oxide films annealed at 573 and 673 K respectively.

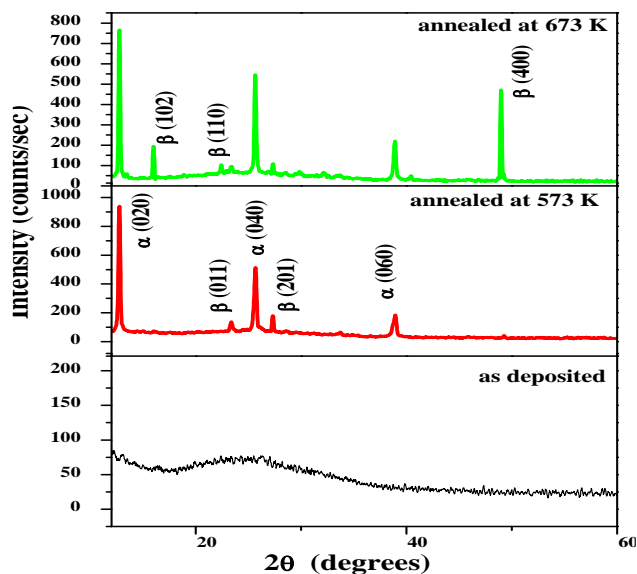


Fig. 2. XRD profiles of as-deposited and annealed MoO₃ film.

Mahammad et al. [27] achieved amorphous films on unheated substrates by DC magnetron sputtering. The films formed at high oxygen partial pressure of 0.4 Pa and annealed at 823 K were completely transformed in to α -phase MoO₃ where as those formed at low oxygen partial pressure of 0.19 Pa showed weaker α - MoO₃ peaks with a weak β - MoO₃ contribution after annealing at 873 K. Fan et al. [28] reported that films deposited at substrate temperatures of 573 and 673 K by RF magnetron sputtering exhibited the growth of β - phase Mo₉O₂₆ and α -phase MoO₃. Yang et al. [29] reported that β - phase MoO₃ films at substrate temperature of 473 K and α - phase at temperatures ≥ 573 K in electron beam evaporated films. Thus, the crystallinity increased with increase of annealing temperature which leads to the growth of nanocrystalline structure of MoO₃. The crystallite size (L) of the films were calculated from the X-ray diffraction peaks by using Debye - Scherrer's relation [30].

$$\text{Crystallite size, } L = 0.89\lambda / \beta \cos\theta \quad (1)$$

where λ is the wavelength of the X-ray (0.15406 nm), β the full width at half maximum of diffraction intensity measured in radians and θ the diffraction angle. The crystallite size of the films increased from 32 and 39 nm with increase of annealing temperature from 573 and 673 K respectively. Thus, crystallite size increased with increase of annealing temperatures as noticed by decreasing in the width of FWHM. This confirms that the amorphous MoO₃ films were modified into nanocrystalline with annealing temperature. The dislocation density (δ) and strain (ϵ) developed in the films were calculated from the crystallite size using the relations [31],

$$\text{Dislocation density, } \delta = 1/L^2 \quad (2)$$

$$\text{and, Strain, } \epsilon = \beta \cos\theta/4 \quad (3)$$

The dislocation density of the films decreased from the 7.6×10^{14} to 6.3×10^{14} line/m² with increase of annealing temperature from 573 to 673 K respectively. The strain developed in the films also decreased from 9.5×10^{-4} to 8.7×10^{-4} line/m² with increase of annealing temperature from 573 to 673 K respectively as shown in **Table 2**. The thermal energy gained by annealing leads to rearrangement of atoms takes place and increase in the crystallite size. The increase of crystallite size leads to decrease in the grain boundary area which will decrease in the dislocation density. When the dislocation density decreased the strain in the films also decreased.

Table 2. FWHM, crystallite size, strain and dislocation density of MoO₃ films.

Temperature	FWHM (Degree)	Crystallite size (nm)	Strain (Line ⁻² m ⁻¹)	Dislocation density (Line/m ²)
As-deposited	---	---	---	---
Annealed at 573 K	0.218	32	9.5×10^{-4}	7.6×10^{14}
Annealed at 673 K	0.202	39	8.7×10^{-4}	6.3×10^{14}

Surface morphology

Fig. 3 shows the scanning electron micrograph images of the as-deposited and the MoO₃ films annealed at different temperatures. SEM of as-deposited films showed smooth surface morphology because of the amorphous phase. The films annealed at 573 K showed the nanoflowers like grain structure with the grain size of 3 μm. At higher annealing temperatures the molecules acquired the required thermal energy to promote the growth of films with larger size grains that is the growth of polycrystalline MoO₃ films with mixed phase. Yang et al. [29] reported on the formation of platelet nanocrystallites uniformly distributed over the surface. In the present work, the films annealed at 673 K were of nanodisk like structure with grain size of about 800 nm, due to the coalescence of neighboring crystallites driven by thermal energy by annealing. The change in surface morphology with different annealing temperatures was due to the modification of grain growth in to nanocrystalline with uniform size of crystallites. The morphological studies revealed that, the annealed films exhibited the layered structure at high annealing temperatures with enhancement in the crystallinity.

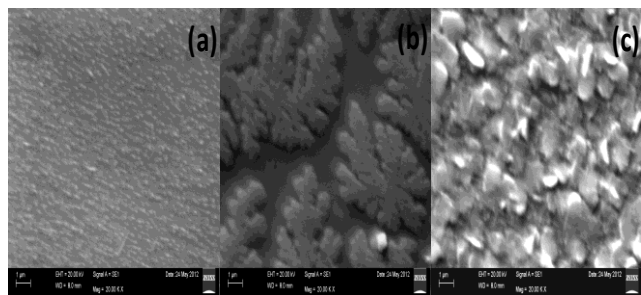


Fig. 3. SEM images of MoO₃ films formed at a) 303 K, and annealed at b) 573 K and c) 673 K.

Fourier transforms infrared spectroscopy

Fourier transform infrared transmittance spectra of as-deposited and annealed MoO₃ films formed on silicon substrates were recorded in the wavenumber range 300 – 1500 cm⁻¹ (**Fig. 4**). The FTIR spectra of as-deposited films contained broad band in between wavenumbers 600 and 1000 cm⁻¹. There were three weak absorption bands at 795, 668 and 570 cm⁻¹ which related to two stretching vibrational modes of Mo=O, Mo-O and one transverse optical vibrations mode of Mo-O-Mo [21]. The FTIR spectra of the films annealed at 573 K showed the intense absorption bands at about 812, 672 and 573 cm⁻¹. The absorption band observed at 812 cm⁻¹ was attributed to the binding vibrations of Mo=O in Mo⁶⁺ of the α-phase MoO₃ [20]. The films annealed at 673 K showed shift in the bands to 817, 681 and 563 cm⁻¹. The intensity of IR absorption bands was increased with increase of annealing temperature. This may be due to enhancement in the ad-atom mobility at high annealing temperatures, hence improved the crystallinity of the films. The intensity of the absorption band at 991 cm⁻¹ was associated with the Mo=O stretching vibration and intensity enhanced with the increase of annealing temperature. Dhanasankar et al. [5] reported that weak absorption peak at 567 cm⁻¹ which represents Mo-O-Mo band and the terminal Mo=O double bands characterized by absorption bands appearing around 885 - 1007 cm⁻¹. Ivanova et al. [32] also noticed, a strong band at 820 cm⁻¹ and a weak absorption band at 545 cm⁻¹, which represent the stretching vibrational modes of Mo=O, and transverse optical vibrations mode of Mo-O-Mo for orthorhombic MoO₃ films annealed at 673 K.

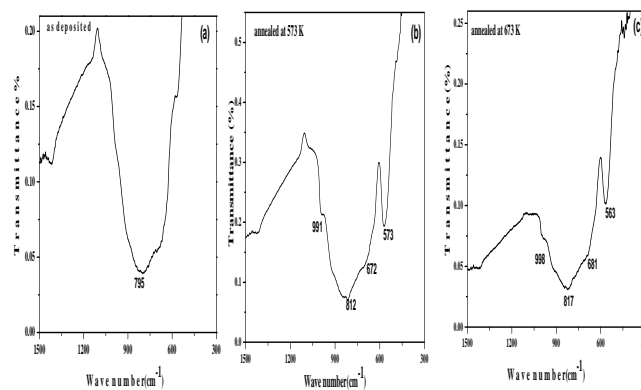


Fig. 4. FTIR spectra of MoO₃ films: (a) as-deposited, and annealed at: (b) 573 K and (c) 673 K.

Optical property

Fig. 5 shows the optical transmittance spectra of as-deposited and annealed MoO₃ films recorded in the wavelength range 300 - 1500 nm. The as-deposited films showed an average transmittance of 60 % (at wavelengths > 600 nm). The films annealed at higher temperatures showed about 85 % transmittance. The low transmittance in the as-deposited films was due to the presence of oxygen vacancies which can scatter the light photons hence the reduced transmittance. The increase of annealing temperature leads to reduction in the oxygen vacancies hence improved crystallinity, hence improved the transmittance. The optical absorption edge of the films

shifted to lower wavelengths side with the increase of annealing temperature.

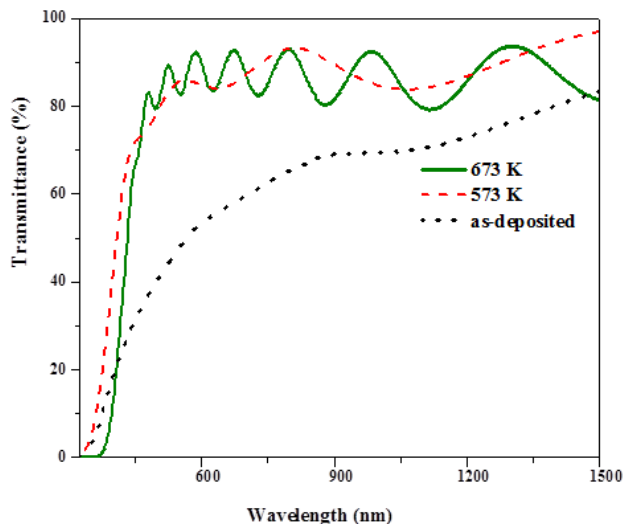


Fig. 5. Optical transmittance spectra of as-deposited and the MoO₃ films annealed at different temperatures.

The optical absorption coefficient (α) of the films was evaluated from the optical transmittance (T) data using the relation,

$$\alpha = - (1/t) \ln T \quad (4)$$

where t is the film thickness. The optical band gap of the films was calculated from the Tauc's [33] plots of $(\alpha h\nu)^2$ versus photon energy ($h\nu$) by assuming direct transition takes place in the films. The optical band gap of the as-deposited films was 2.98 eV. The band gap increased to 3.08 and 3.15 eV in the films annealed at 573 and 673 K respectively. As the annealing temperature increased the optical band gap of the film increased. Sivakumar et al. [25] achieved low optical band gap of 2.8 eV in the MoO₃ films formed at room temperature, while those deposited at 473 K showed lower band gap of 2.3 eV. In the literature, Navas et al. [21] reported that the optical band gap values of MoO₃ films were 3.77, 3.82 and 3.87 eV for as-deposited and the films annealed at 573 and 673 K respectively. Increase in band gap with annealing temperature can be attributed to the partial filling of oxygen vacancies. The band gap values are in consistent with the value of 3.16 and 3.27 eV obtained in electron beam evaporated films by Sain and reddy [15] for as-deposited and annealed MoO₃ films respectively.

The refractive index (n) of the films was determined from the optical transmittance interference data employing Swanepoel's envelope method employing the relation [34],

$$n(\lambda) = [N + (N^2 - n_0^2 n_1^2)^{1/2}]^{1/2} \quad (5)$$

$$N = 2n_0 n_1 [(T_M - T_m) / T_M T_m] + [(n_0^2 + n_1^2) / 2] \quad (6)$$

where T_M and T_m are the optical transmittance maxima and minima and n_0 and n_1 the refractive indices of air and substrate respectively. The variation of refractive index

with wavelength was shown in Fig. 6. The refractive index decreased with increase of annealing temperature. The refractive index (at wavelength of 600 nm) of the as-deposited film was 2.01 and increased to 2.08 with increase of annealing temperature to 673 K due to improvement in the crystallinity and packing density of the films. In contrast to this, Mohamed and Venkataraj [27] observed that the refractive index decreased with increase of annealing temperature up to 723 K, in the films formed at different sputter pressures, but after 773 K the refractive index increased. Navas et al. [35] noticed that the refractive index of as-deposited films was 1.60 and decreased to 1.36 with the increase of annealing temperature to 673 K.

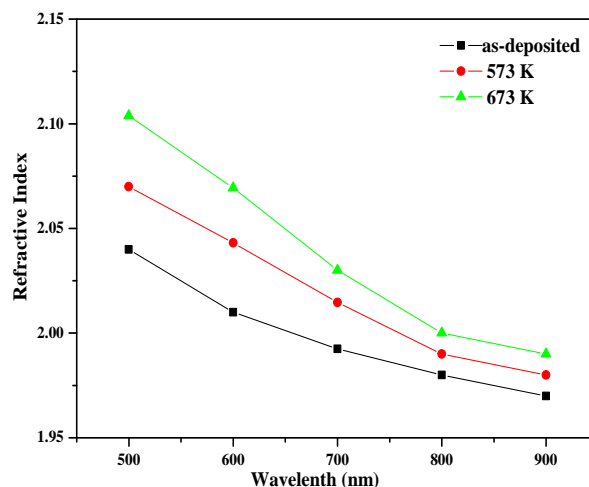


Fig. 6. Variation of refractive index with wavelength of as-deposited and annealed MoO₃ film.

Conclusion

Thin MoO₃ films were formed on Corning glass and silicon substrates held at room temperatures by sputtering of molybdenum target in an oxygen partial pressure of 4×10^{-4} mbar. The as-deposited films were annealed in air for two hours at temperatures in the range 573 - 673 K. X-ray diffraction studies revealed that the as-deposited films were amorphous in nature. Post-deposition annealing of films at 573 K was transformed into nanocrystalline with mixed phase of orthorhombic and monoclinic MoO₃. Further increase of annealing temperature to 673 K the crystallinity of the films improved. The crystallite size of the films increased from 32 to 39 nm with the increase of annealing temperature from 573 to 673 K respectively. The Fourier transform infrared transmittance spectra of the films indicated that the presence of characteristic vibrational modes of Mo = O and Mo - O - Mo related to the growth of MoO₃. Scanning electron micrographs of as-deposited films shows fine grain structure, nanoflower like structure growth in the films annealed at 573 K and nanodisk like structures at higher temperature of 673 K. The optical transmittance of the as-deposited films was about 60 % and it increased to 85 % with increase of annealing temperature to 673 K. The optical absorption edge of the films shifted towards lower wavelength side with increase of annealing temperatures. The optical band gap of the as-deposited films was 2.98 eV and it increased to 3.15 eV in the films annealed at 673 K. The refractive index of the as-deposited

film was 2.01 and increased to 2.08 with increase of annealing temperature to 673 K due to improvement in the crystallinity and packing density of the films.

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