www.vbripress.com, www.amlett.com, DOI: 10.5185/amlett.2010.11176

Published online by the VBRI press in 2011

X-ray and electrical properties of Ba(Gd_{0.5}Nb_{0.5})O₃ ceramic

Pritam Kumar¹, B.P. Singh², T.P. Sinha³, N.K. Singh^{1,*}

Received: 11 Nov 2010, Revised: 9 Dec 2010 and Accepted: 12 Dec 2010

ABSTRACT

The complex perovskite oxide barium gadolinium niobate, $Ba(Gd_{0.5}Nb_{0.5})O_3$ (BGN) is synthesized by a solid-state reaction technique. The X-ray diffraction of the sample at room temperature shows a tetragonal phase. The scanning electron micrograph of the sample shows the average grain size of BGN $\sim 1.92~\mu m$. The frequency-dependent dielectric dispersion of BGN is investigated in the temperature range from 303 K to 463 K and in a frequency range from 50 Hz to 1 MHz by impedance spectroscopy. The logarithmic angular frequency dependence of loss peak is found to obey an Arrhenius law with activation energy of 0.70 eV. It is observed that: (i) the dielectric constant (ϵ') and loss tangent ($\tan \delta$) are dependent on frequency, (ii) the temperature of dielectric constant maximum shift toward lower temperature side, (iii) The scaling behavior of dielectric loss spectra suggests that the relaxation describes the same mechanism at various temperatures. Copyright © 2011 VBRI press.

Keywords: Dielectrics; electrical properties; X-ray diffraction; scanning electron micrographs.



Pritam Kumar obtained M.Sc. degree in physics from T. M. Bhagalpur University, Bhagalpur, India in 2005. He has registered for Ph.D. at T. M. Bhagalpur University on Structural and Dielectric study of some complex perovskite oxides. Presently, he is working as JRF in Major Research Project sponsored by DRDO, New Delhi since 2009 at V.K.S. University, Department of Physics. During his research career, he involved in the preparation of fine ceramics, ferroelectric, piezoelectric and non-lead based materials.

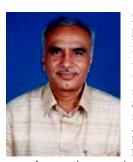


B. P. Singh obtained his M.Sc. degree in physics from T. M. Bhagalpur University, Bhagalpur, Bihar, India in 1980 and Ph.D. in 1989 from T. M. Bhagalpur University, Bhagalpur, Bihar, India. Dr. Singh is presently working at University Department of Physics, T. M. Bhagalpur University, Bhagalpur, Bihar as a Reader/Associate Professor. Dr. Singh is associated with about one dozen scientific organization. His recent field of research is Material Science (Experimental as well as theoretical). He has published over 40 research papers in reputed journals of the world.



T. P. Sinha is Professor of Physics in Bose institute, Kolkata (India). He obtained his M.Sc. and Ph.D. degree in Physics from T. M. Bhagalpur University, Bhagalpur (India). He did his post-doc at Saha Institute of Nuclear Physics, Kolkata (India). He has more than

100 publications in peer reviewed Journals. His major research interests are magnetic, optical and electrical properties of perovskite oxides, electronic band structure calculation, polymers and nanoparticles.



N.K. Singh obtained his M.Sc. degree in physics from Bihar University, Muzaffarpur, India in 1969 and Ph.D. in 1984 from Patna University, Patna. He is Professor and Head, University Department of Physics, V.K.S. University, Ara (India). He is the life member of different academic and Research Societies of India and abroad. He is a member of the Editorial Board of National Journal "ARJP". More than sixteen students have completed their Ph.D. degrees, under his guidance. He has published over 55

research papers in peer reviewed journals of the world. He was a U.G.C. Visiting Associate at I.I.T. Kharagpur, India during 2001-2003. In span of 27 years of his research career, he has remained engaged in the preparation of fine ceramics, ferroelectric piezoelectric and non Lead based materials. Presently, he is Principal Investigator of Major Research Project sponsored by DRDO, New Delhi and is engaged in synthesis and characterization of Lead free ferroelectric-piezoelectric systems for sensor applications.

¹University Department of Physics, V. K. S. University, Ara 802301, India

²University Department of Physics, T. M. Bhagalpur University, Bhagalpur 812007, India

³Department of Physics, Bose Institute, 93/1, A.P.C. Road, Kolkata 700009, India

^{*}Corresponding author. E-mail: singh_nk_phy27@yahoo.com

Introduction

Perovskite materials with high dielectric constants have been used for technological applications such as wireless communication system, cellular phones and global positioning systems in the form of capacitors resonators and filters. High dielectric constant permits smaller capacitive components, thus enabling smaller size of electronic devices [1]. Complex perovskites are very promising for electroceramic applications and many researchers have shown considerable interest in the dielectric properties of these compounds. Materials having a diffuse phase transition (relaxors) have attracted the most attention due to their broad maximum in the temperature dependence of their dielectric constant [2, 3]. The perovskite family is created by the doping of other types of cations into the stoichiometry and / or introducing anion deficiency. In the ABO₃ structure, the valences of the A (12-cordinated) and B (6-cordinated) cations are usually 2⁺ and 4⁺, respectively. During the intervening years, a large amount a good understanding of the many inter-related properties and phenomena associated with these materials has been achieved.

Recently, giant dielectric constant and dielectric relaxation in $A(Fe_{0.5}B_{0.5}O_3)$, (A = Ba, Sr and B = Nb, Ta) [4-7], CaCuTi₄O₁₂ [8], CdCr₂S₄ [9] have been pursued to understand the relaxation mechanism, which describes the dielectric relaxation i.e. charge redistribution, structural frustration or polaron redistribution, ferroelectric relaxor, and Maxwell-Wangner space charge (pseudo relaxor). The high value of dielectric constant over a very wide temperature interval is due to disorder in the distribution of B-site ions in the perovskite unit cell. This may lead to composition fluctuations and, as a consequence, to different local curie temperatures in the different regions of the ceramic. Using Fourier transform infrared spectroscopy and analyzing the reflectivity spectra, an investigation of the polar phonons of Ba(B'_{1/2}B"_{1/2})O₃ ceramics with B' = Nd³⁺ Gd^{3+} , Y^{3+} , In^{3+} , Cd^{2+} or Mg^{2+} and $B'' = Ta^{5+}$, Nb^{5+} or W^{6+} was carried by Zurmuhlen et al. to find a correlation between ionic parameters of ceramic materials and their complex permittivity at microwave frequency [10]. Phase transition and microwave dielectric properties in Ca(Al_{1/2}Nb_{1/2})O₃ and its solid solution with CaTiO₃ have been analyzed by Levin et al. using X-ray and neutron power diffraction, transmission electron microscope, Raman spectroscopy and dielectric measurements [11].

In this paper we investigate the structural and electrical impedance study of the barium gadolinium niobate, $Ba(Gd_{0.5}Nb_{0.5})O_3$ (BGN) ceramic in the temperature range from 323 K to 463 K and in the frequency range from 50 Hz to 1 MHz by means of dielectric spectroscopy. Dielectric spectroscopy allows measurement of the capacitance (C) and tangent loss ($\tan \delta$) over a frequency range at various temperatures. From the measured capacitance (C) and tangent loss ($\tan \delta$), four complex dielectric functions assumed can be computed: impedance (Z*), permittivity (ϵ *), electric modulus (M*) and admittance (Y*). Studying electrical data in the different functions allow different features of the materials to be recognized. Study of electrical properties, such as dielectric constant, loss tangent, AC conductivity, etc. in this oxide over wide range of frequency and temperature will help

us in assessing its insulating character for potential application.

Experimental

Materials and Methods

The solid state reaction technique was employed for the synthesis of BGN. The raw chemicals used in this study were reagent grade BaCO₃, Gd₂O₃, and Nb₂O₅. The mixed raw chemicals in stoichiometric ratio were calcined in alumina crucible at 1300 °C in air for 10 h and brought to room temperature under controlled cooling. The calcined sample was palletized into disc using polyvinyl alcohol as binder. Finally, the discs were sintered at optimized temperature 1100°C for 10 h. The X-ray powder diffraction pattern of the sample at room temperature and the scanning electron micrograph (inset) are shown in Fig. 1. All the reflection peaks of the X-ray profile were indexed and lattice parameters were determined using least-squares method with the help of standard computer program (crys fire). A good agreement between the observed and calculated interplaner spacing (d-values) suggests that the compound has a tetragonal structure at room temperature. The least-squares refined cell parameters for tetragonal unit cell are: a = 2.117 Å, c = 3.001 Å. The scanning electron micrograph of the sample was recorded by FEI Quanta 200 equipment to check proper compactness of the sample. From inset of Fig.1, the nature of the micrographs exhibits the polycrystalline texture of the material having highly distinctive and compact rectangular/cubical distributions (with less voids). The grain size of the pellet sample was found to be 1.92 µm. Careful examinations (scanning) of the complete surface of the sample exhibits that the grains are homogeneously distributed throughout the surface of the sample. The flat polished surface of sintered pellets were electroded with air drying silver paste and fired at 423 K for 2h before taking any electrical measurement. From the measurement, we have obtained capacitance (C) and tangent loss (tand) by using an LCR meter (Hioki) in the temperature range from 303K to 463K and in the frequency 50Hz to 1 MHz; using (C) and (tan δ), we have computed dielectric constant (ε') and conductivity (σ) as follows:

$$\varepsilon' = \varepsilon_0 (C/C_0) - (1)$$

$$\sigma = \omega \varepsilon_0 \varepsilon'' - (2)$$

where \mathcal{E}_0 is the dielectric permittivity in air, C/C_0 the ratio of capacitance measured with dielectric and without dielectric, ω the angular frequency and $\mathcal{E}'' = \tan \delta \, \varepsilon'$.

Results and discussion

Permittivity analysis

The relation of angular frequency, ω (= $2\pi v$) with dielectric constant (ε') and dielectric loss (tan δ) at various temperatures for BGN is described in **Fig. 2**. A relaxation is observed as a gradual decrease in $\varepsilon'(\omega)$ and as a broad peak in tan δ (ω). The increasing value of ε' at frequencies below 500 Hz may arise due to space charge polarization.

This behavior is also found in other compounds of the family studied by us in their ceramic form [12-15]. It is evident from Fig. 2(b) that tan δ peaks (tan δ_{max}) shifts to higher frequency with increasing temperature and that a strong dispersion of tan δ exists in BGN [16].

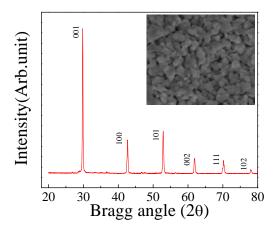


Fig. 1. XRD pattern and scanning electron micrograph (inset) of the $Ba(Gd_{0.5}Nb_{0.5})O_3$ at room temperature.

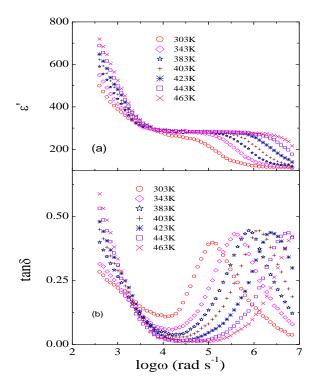


Fig. 2. Logarithmic angular frequency dependence of ϵ' (a) and $tan\delta$ (b) of $Ba(Gd_{0.5}Nb_{0.5})O_3$ at various temperatures.

The most probable relaxation time follows the Arrhenius law given by:

$$\omega_m = \omega_0 \exp\left[\frac{-E_a}{k_B T}\right]$$
 -----(3)

where ω_0 is the pre-exponential factor and E_a is the activation energy. A plot of $\log \omega$ vs. 1/T is shown in **Fig.** 3, where the symbols are the experimental data points and the solid line is the least squares straight line fit to the data. The activation energy calculated from least square fit to the points is 0.70 eV. It seems clear that the width of loss peaks in **Fig.** 2(b) cannot be accounted for in terms of monodispersive relaxation, but points towards the possibility of a distribution of relaxation times.

If we plot the tan δ data in scaled coordinates, i.e. tan δ /tan δ_m and log (ω/ω_m) , where ω_m corresponds to the frequency of the loss peak in the tan δ vs. log ω plots, the entire dielectric loss data can be collapsed into one master curve as shown in **Fig. 4**. The scaling behavior of tan δ clearly indicates that the relaxation follows the same mechanism at various temperatures.

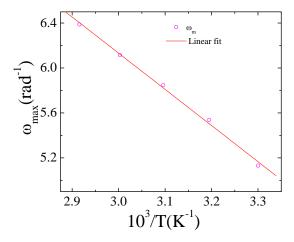


Fig. 3. The Arrhenius plot of ω_m corresponding to $\tan \delta$, where symbols are the experimental points and the solid line is the least squares fit to the experimental data.

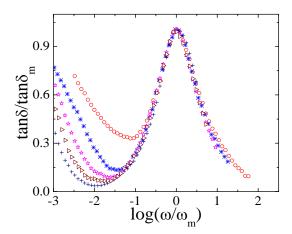


Fig. 4. Scaling behavior of $tan\delta$ at various temperatures for $Ba(Gd_{0.5}Nb_{0.5})O_3$.

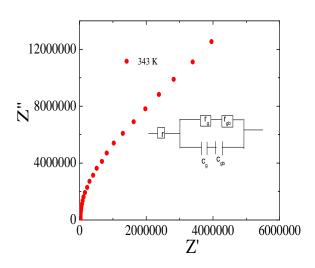


Fig. 5. Complex plane impedance plot of Ba(Gd_{0.5}Nb_{0.5})O₃ at 343 K, and inset shows RC equivalent circuit.

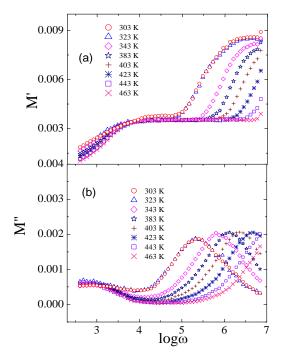


Fig. 6. Logarithmic angular frequency dependence of M^\prime (a) and $M^{\prime\prime}$ (b) of $Ba(Gd_{0.5}Nb_{0.5})O_3$ at various temperatures.

Modulus analysis

Complex plane impedance plots of BGN, plotting the imaginary part Z" against the real part Z' at temperature 343 K is shown in Fig. 5. For a single phase polycrystalline dielectric material having interfacial boundary layers (grain-boundary), two semicircular arcs or depressed circular arcs may be obtained in the complex plane plot, which can be explained as two parallel RC elements connected in series (inset of Fig. 5), one branch is

associated with the grain and other with the grain-boundary of the sample. The real (Z') and imaginary (Z'') of total impedance of the equivalent circuit are defined as:

$$Z' = R + \frac{R_g}{1 + (\omega R_g C_g)^2} + \frac{R_{gb}}{1 + (\omega R_{gb} C_{gb})^2}$$
-----(4)

$$Z'' = R_g \left[\frac{\omega R_g C_g}{1 + (\omega R_g C_g)^2} \right] + R_{gb} \left[\frac{\omega R_{gb} C_{gb}}{1 + (\omega R_{gb} C_{gb})^2} \right]$$
-----(5)

where R_g , C_g are the grain resistance and capacitance and R_{gb} , C_{gb} are the corresponding quantities for grain-boundary.

Modulus analysis

Complex modulus, electric modulus or inverse complex permittivity, M*, is defined by the following equation

$$M^{*}(\omega) = \frac{1}{\varepsilon^{*}} = \frac{1}{\varepsilon' - j\varepsilon''} = \frac{\varepsilon'}{\varepsilon'^{2} + \varepsilon''^{2}} + j\frac{\varepsilon'}{\varepsilon'^{2} + \varepsilon''^{2}} = M'(\omega) + jM''(\omega)$$

$$= M_{\infty} [1 - \int_{0}^{\infty} e^{-j\omega t} (\frac{d\phi(t)}{dt}) dt] \qquad ------(6)$$

where M' is the real and M" the imaginary electric modulus, and ε' the real and ε'' the imaginary permittivity and $\phi(t)$ gives the time evolution of the electric field within the dielectric.

Fig. 6 displays the angular frequency dependence of $M'(\omega)$ and $M''(\omega)$ for BGN as a function of temperature. $M'(\omega)$ shows a dispersion tending towards M_{∞} (the asymptotic value of $M'(\omega)$ at higher frequencies in Fig. 6(a), while $M''(\omega)$ exhibits a maximum (M''_m) centered at the dispersion region of $M'(\omega)$. In Fig. 6(b) the position of the peak M''_m shifts to higher frequencies as the temperature is increased. The frequency region below peak maximum M'' determines the range in which charge carriers are mobile on long distances. At frequency above peak maximum M'' the carriers are confined to potential wells, being mobile on short distances.

Conductivity analysis

If we assumes that all dielectric loss in the temperature range studied are due to conductivity, the conductivity can be expressed as $\sigma\left(\omega\right)=\omega\epsilon_{0}\epsilon''$, here σ is the real part of the conductivity and ϵ'' is the imaginary part of dielectric constant. The frequency spectra of the conductivity for

BGN at different measuring temperatures are shown in **Fig.** 7. The conductivity shows dispersion which shifts to higher frequency side with the increase of temperature. It is seen from Fig. 7 that σ decreases with decreasing frequency and becomes independent of frequency after a certain value. Extrapolation of this part towards lower frequency will give $\sigma_{dc.}$

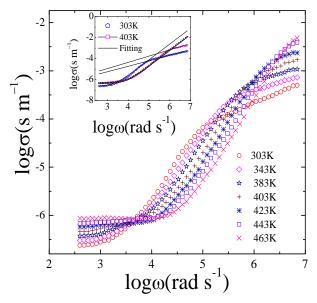


Fig. 7. Logarithmic angular frequency dependence of conductivity (σ) for Ba(Gd_{0.5}Nb_{0.5})O_{3 3} at various temperatures. A fitting by power law for the same at 303 and 403 K is shown by solid lines in the inset.

The very basic fact about AC conductivity in BGN is that σ is an increasing function of frequency. The real parts of conductivity spectra can be explained by the power law define as [17, 18]:

$$\sigma = \sigma_{dc} \left[1 + \left(\frac{\omega}{\omega_H} \right)^n \right] \qquad (5)$$

where σ_{dc} is the DC conductivity, ω_H is the hopping frequency of the charge carriers, and n is the dimensionless frequency exponent. The experimental conductivity spectra of BGN are fitted to Eq. (7) with σ_{dc} as variable keeping in mind that the value of parameter n are weakly temperature dependent. The best fit of the conductivity spectra are shown in inset of **Fig. 7** by solid lines.

Conclusion

Logarithmic angular frequency dependence of dielectric constant and loss tangent of the barium gadolinium niobate, Ba(Gd_{0.5}Nb_{0.5})O₃ (BGN) ceramic synthesized by a solid state reaction technique is investigated in the temperature range from 303 to 463 K. The X-ray diffraction of the sample at room temperature shows a tetragonal phase. The scanning electron micrograph of the sample shows the average grain size of BGN ~1.92 µm. The relaxation mechanism has been discussed in the framework of electric modulus. The logarithmic angular frequency dependence of the loss peak is found to obey an Arrhenius law with

activation energy of 0.70 eV. The scaling behavior of the loss tangent suggests that the relaxation mechanism describes the same mechanism at various temperatures.

Acknowledgement

The authors are grateful to Prof. R.N.P. Choudhary, I.I.T Kharagpur for his help and suggestions.

References

Cava, R. J. J. Mater. Chem. 2001, 11, 54.

DOI: 10.1039/B003681L

Intatha, U.; Eitssayeam, S.; Pengpat, K.; MacKenzie, J.D.; Kenneth, T. T. Materials Letters 2007, 61, 196. **DOI:** 10.1016/j.matlet.2006.04.030

3. Roulland, F.; Terras, R.; Allainmat, G.; Pollet, M.; Marinel, S. J. Eur. Ceram. Soc. 2004, 24, 1019.

DOI: <u>10.1016/S0955-2219(03)00553-3</u>

Liu, Y.Y.; Chen, X. M.; Liu, X.; Li, L. Appl. Phys. Lett. 2007, 90, 192905.

DOI:10.1063/1.2737905

Chung, C. Y.; Chung, Y. H.; Chen, G. J. J. Appl. Phys. 2004, 96,

DOI: <u>10.1063/1.1804243</u>

Saha, S.; Sinha, T. P. J. Phys. Condens Matter 2002, 14, 249.

DOI: 10.1088/0953-8984/14/2/31

Wang, Z.; Chen, X. M.; Ni, L.; Liu, Y.Y.; Liu, X. Q. Appl. Phys. Lett. **2007**, 90, 102905.

DOI: <u>10.1063/1.2711767</u>

Sinclair, D. C.; Adams, T. B.; Morrison, F. D.; West, A. R. Appl. Phys. Lett. 2002, 80, 2153.

DOI: <u>10.1063/1.1463211</u>

Catalan, G. Appl. Phys. Lett. 2006, 88, 102902.

DOI: 10.1063/1.2177543

10. Zurmuhlen, R.; Petzett, J.; Kamba, S.; Voitsekhovskii, V. V.; Colla, Setter, E.; N. J. Appl. Phys. 1995, 77, 5341. DOI: 10.1063/1.359597

11. Levin, I.; Chen, J. Y.; Maslar, J. E.; Vanderah, T. A. J. Appl. Phys. 2001, 90, 904.

DOI: <u>10.1063/1.1373705</u>

12. Singh, N. K.; Choudhary, R. N. P.; Banarji, Behera Physica B 2008, 403, 1673.

DOI: <u>10.1016/j.physb.2007.09.083</u>

13. Singh, N. K.; Kumar, P.; Kumar, H.; Rai, R. Adv. Mat. Lett. 2010, 1,

DOI: 10.5185/amlett.2010.3102

14. Singh, N. K.; Kumar, P.; Roy, O. P.; Rai, R J. Alloys and Compounds 2010, 507, 542.

DOI: <u>10.1016/j.jallcom.2010.08.015</u>

15. Kumar, P.; Singh, B. P.; Sinha, T. P.; Singh, N. K. Physica B 2011, 406, 139.

DOI: 10.1016/j.physb.2010.09.019

16. Dutta, Alo; Sinha, T. P. Phys. Rev. B 2007, 76, 155113. DOI: 10.1103/PhysRevB.76.155113

Almond, D. P.; West, A. R. Nature 1983, 306, 453.

DOI: 10.1038/306456a0

18. Hairetdinov, E. F.; Uvarov, N. F.; Patel, H. K.; Martin, S. W. Phys. Rev. B 1994, 50, 13259.

DOI: <u>10.1103/PhysRevB.50.13259</u>

ADVANCED MATERIALS Letters

Publish your article in this journal

ADVANCED MATERIALS Letters is an international journal published quarterly. The journal is intended to provide top-quality peer-reviewed research papers in the fascinating field of materials science particularly in the area of structure, synthesis and processing, characterization, advanced-state properties, and applications of materials. All articles are indexed on various databases including DOAJ and are available for download for free. The manuscript management system is completely electronic and has fast and fair peer-review process. The journal includes review articles, research articles, notes, letter to editor and short communications.

Submit your manuscript: http://amlett.com/submitanarticle.php

